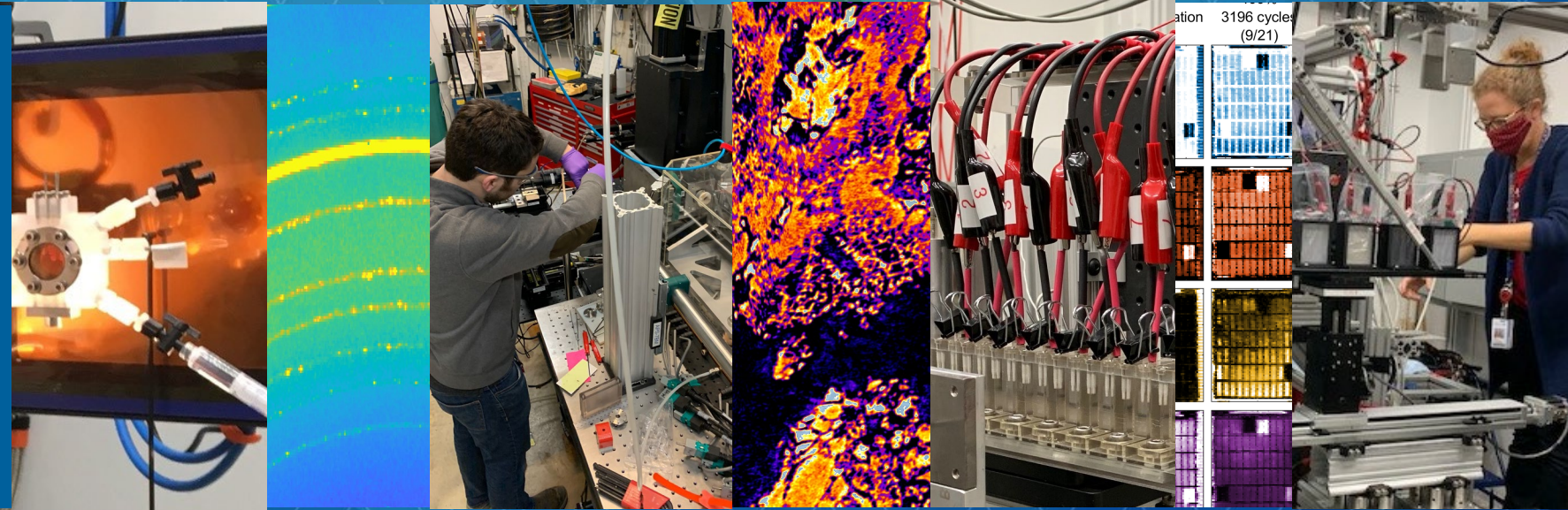


BCI/MPSC Lead Battery Symposium
December 15, 2022

IMAGING CHARGE ACCEPTANCE IN LEAD BATTERIES: A MULTISCALE CHALLENGE

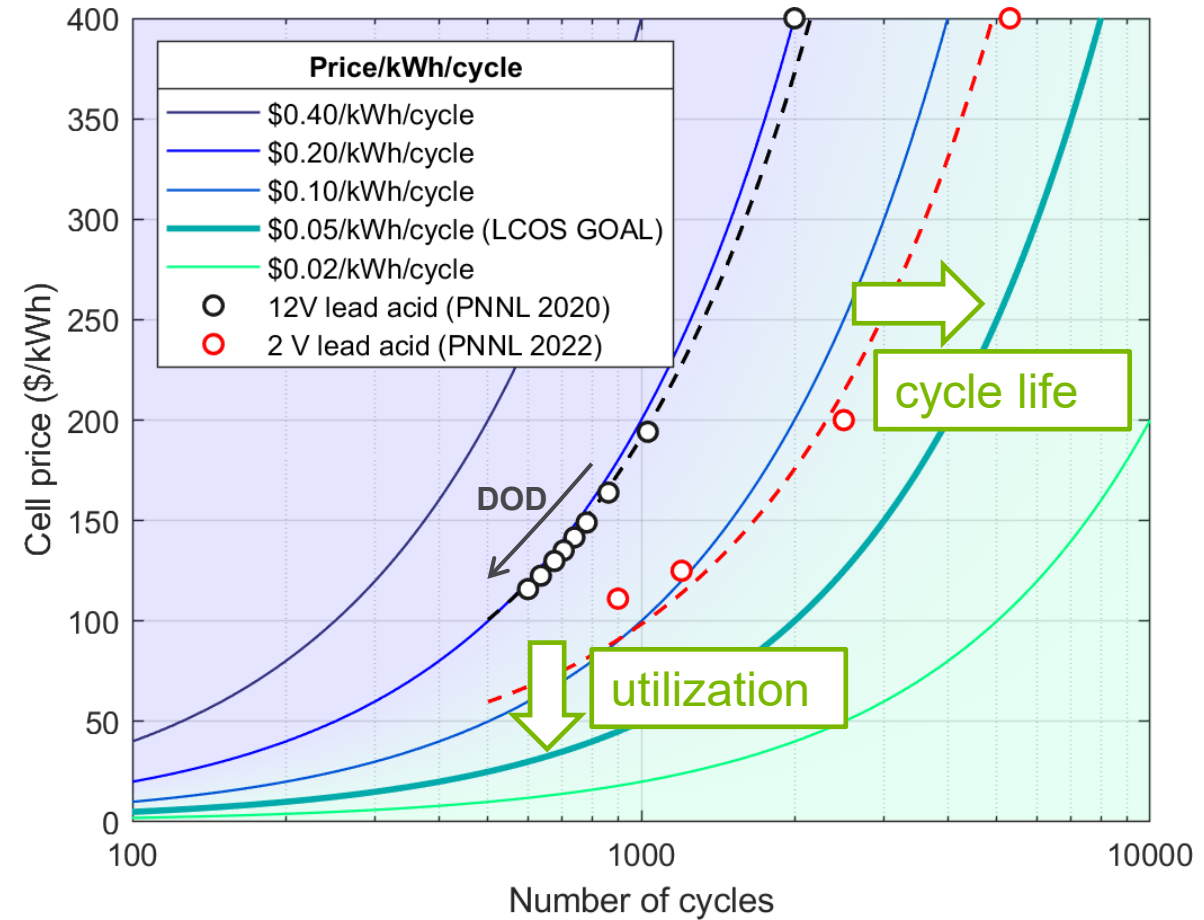


TIM FISTER, TIFFANY KINNIBRUGH, MARK WOLFMAN, JUAN GARCIA, HAKIM IDDIR,
KEVIN KNEHR, MOHAMMED EFFAT, ANA SUZANA, PIETRO PAPA-LOPES
Argonne National Laboratory

MOTIVATION

Improving lead acid for pulse power *and* deep cycle applications

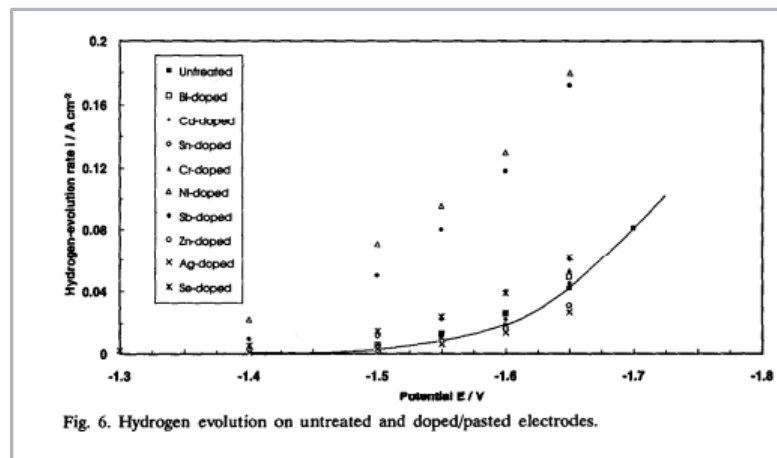
- Lead batteries are domestically manufactured, >99% recycled, and use inexpensive materials.
- At the **cell level**, current lead acid cells already approach LCOS goals highlighted in the Energy Storage Grand Challenge.
- And despite being a mature technology, there is still significant **room for growth** in utilization and cycle life in lead acid.



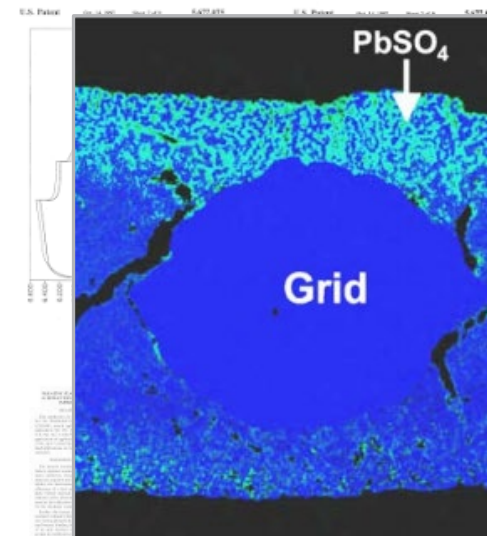
MOTIVATION

Is there anything left to study?

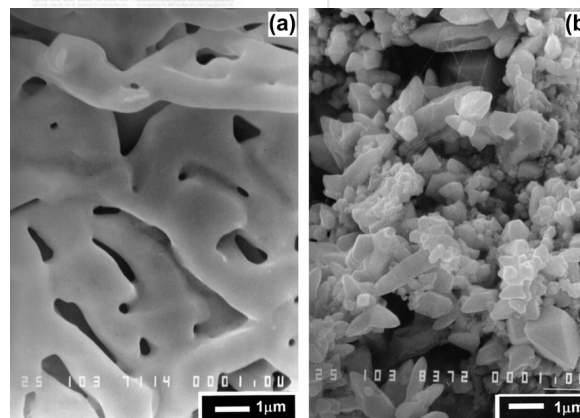
- 150 years of heuristic knowledge, largely around:
 - Cell and grid design, modeling
 - Control of dopants, additives
 - Influence of morphology, porosity
- Lead batteries have largely relied on cycling data or *ex situ* data for evaluation.
- Tools unique to national labs can provide new *operando* information...



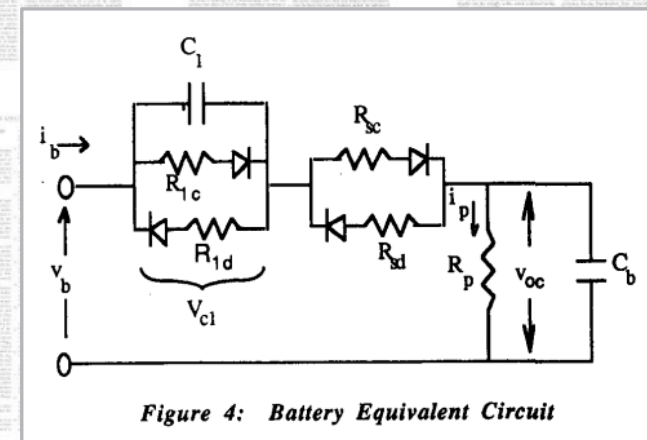
Lam et al JPS 1994



Lam et al JPS 2004



Iliev and Pavlov, J. App. Electrochem. 1985



Salameh IEEE 1992

EXAMPLE

Advanced Photon Source (APS)

Synchrotrons: cutting edge x-ray sources capable of resolving complexity over many length scales.

- Higher energy = *in situ*
- Higher flux = *real time*
- Higher resolution = *resolve heterogeneity* (such as chemical gradients, particles)

APS is a *billion-dollar* instrument, the type of user facility made possible by the DOE national labs.



IMPROVING UTILIZATION AND CYCLE LIFE

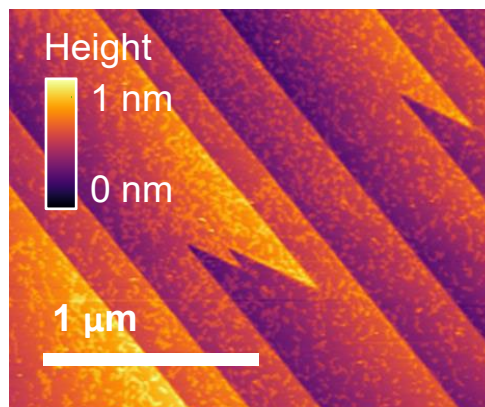
A challenge over many length scales

Improving utilization and lifetime is a multiscale problem involving **solid** and **liquid** species.

- Atomic level issues: PbSO_4 nucleation, Pb^{2+} solvation, acid dissociation, additives/dopants
- Particle level issues: sulfation/pore clogging, diffusion/tortuosity limitations for e^- and SO_4^{2-}
- Cell level issues: electrolyte stratification, grid corrosion, paste shedding/softening
- Battery level issues: charge/discharge procedures, impedance, electrolyte management, grid design

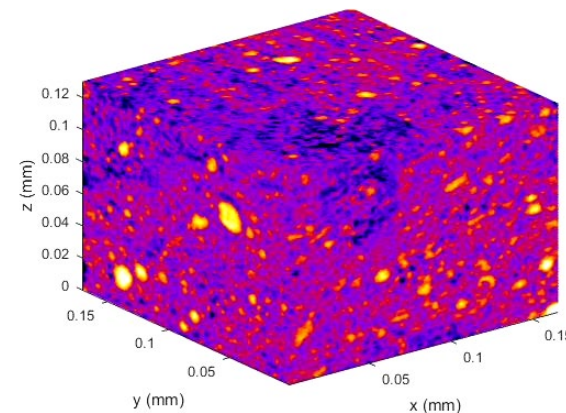
Atomic level (~nm, nAh)

Example: monolayer growth of PbSO_4 on barite 001.



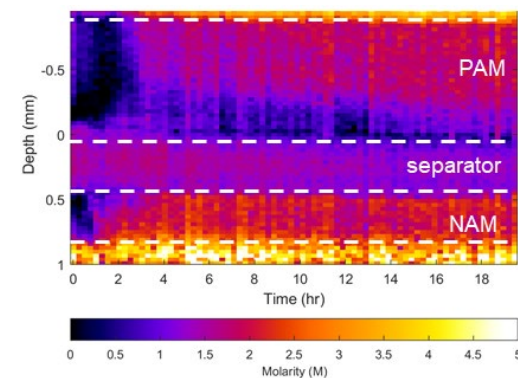
Particle level (~ μm , μAh)

Example: Reconstructed volume from CT scan of a paste electrode (color = density)



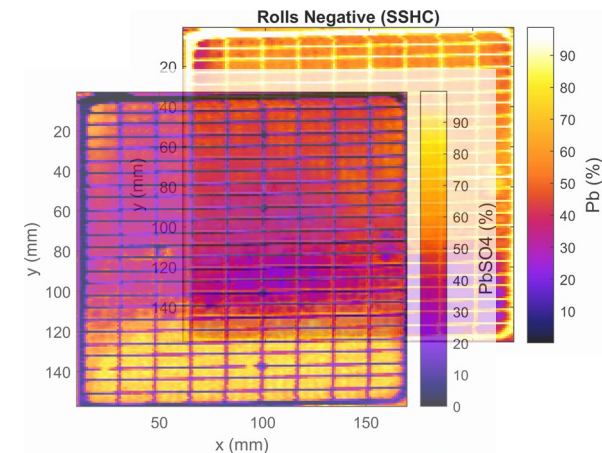
Cell level (~mm, mAh-Ah)

Example: Electrolyte mapping during formation using high energy x-ray scattering



Battery level (~100 Ah)

Example: cycling at PNNL and XRD from EOF battery plates



CURRENT PROGRAMS

Analysis from fundamental to applied

Model expanders

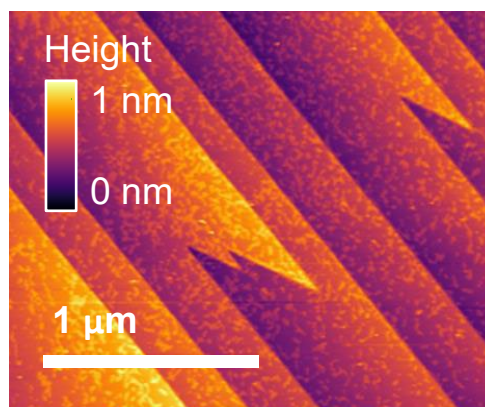
PIs: Papa Lopes, Lind-Kovacs

DOE/OE:

fundamental science
(PIs: Murugesan, Fister)

Atomic level (~nm, nAh)

Example: monolayer growth of PbSO_4 on barite 001.

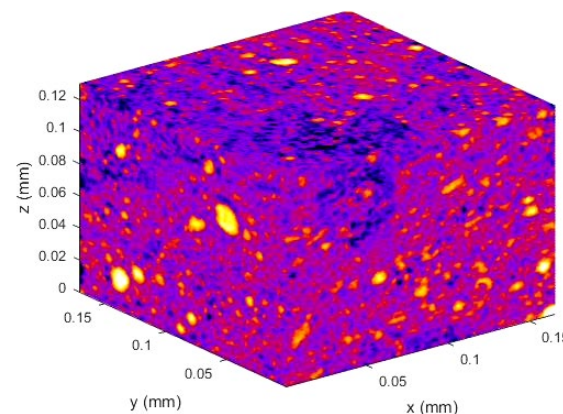


LBSRP

(Fister, Papa Lopes)

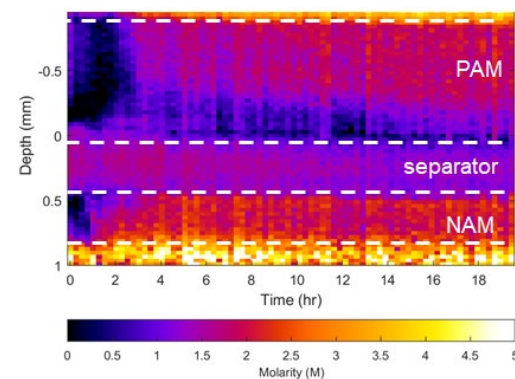
Particle level (~ μm , μAh)

Example: Reconstructed volume from CT scan of a paste electrode (color = density)



Cell level (~mm, mAh-Ah)

Example: Electrolyte mapping during formation

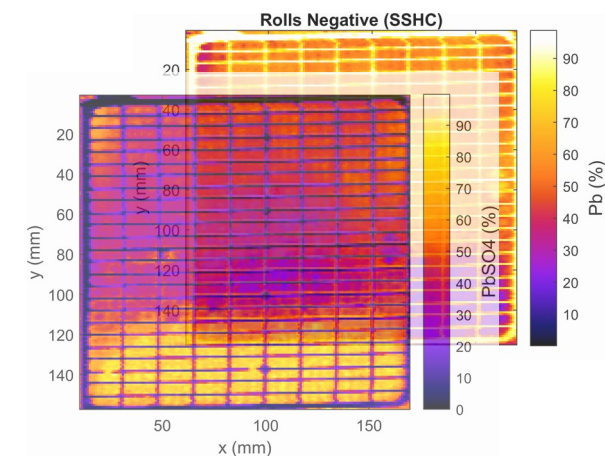


DOE/OE:

Testing and evaluation
(PI: Thomsen)

Battery level (~100 Ah)

Example: cycling at PNNL and XRD from EOF battery plates



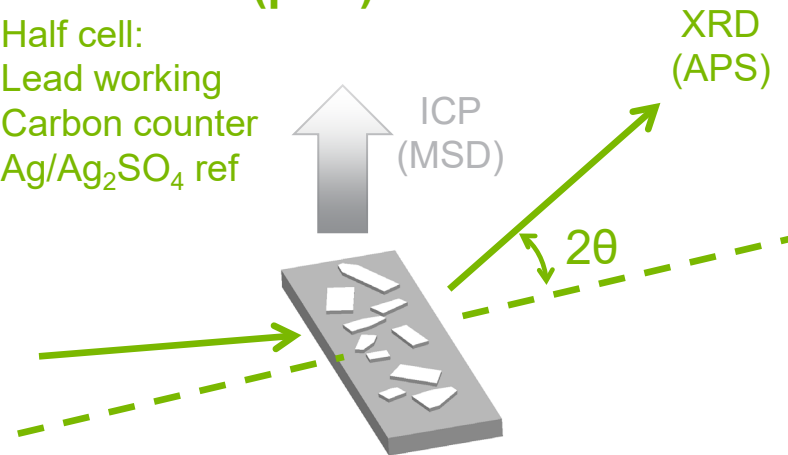
EXPERIMENTAL SETUPS

Increasing complexity

- Initial emphasis was on [simpler] processes taking place on negative electrode.
- Transitioned to pasted cells, but maintained emphasis on negative additives.
- Final phase (on-going): developing robust two-electrode cell to study failure mechanisms.

Plante cell (μAh)

Half cell:
Lead working
Carbon counter
Ag/Ag₂SO₄ ref

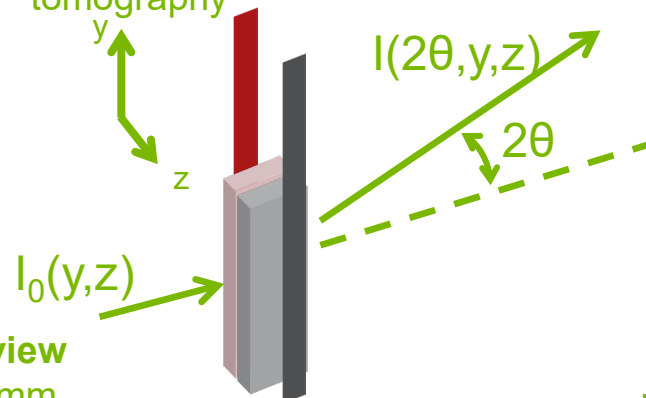


Grazing incidence
Beam = 1x3 mm²
Sample = 8x3 mm²

XRD
(APS)

Mini cells (mAh):

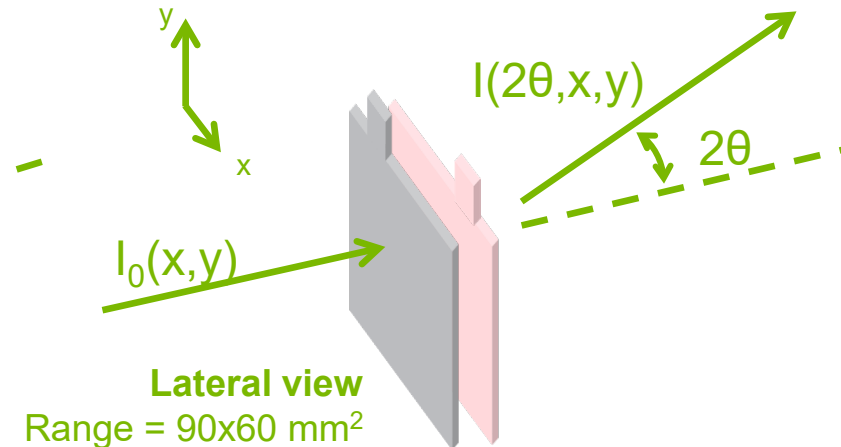
Depth profiling and computed tomography



Edge view
Range: 3 mm,
 $\Delta z = 0.03$ mm

2V pasted cells (Ah)

2 electrode cells



Lateral view
Range = 90x60 mm²
 $\Delta xy = 1$ mm²

PLANTE CELLS

X-ray scattering from a lead surface

- Example: APS, 13BMC
- Energy = 28 keV ($\lambda = 0.43 \text{ \AA}$)

A photograph of an X-ray scattering experimental setup. The setup includes a sample cell, beam-defining optics, a detector, and a potentiostat. Orange arrows point from labels to the corresponding components. A dashed orange line indicates the path of the X-ray beam from the sample to the detector.

Beam defining optics, slits, detectors

Sample
(echem cell)

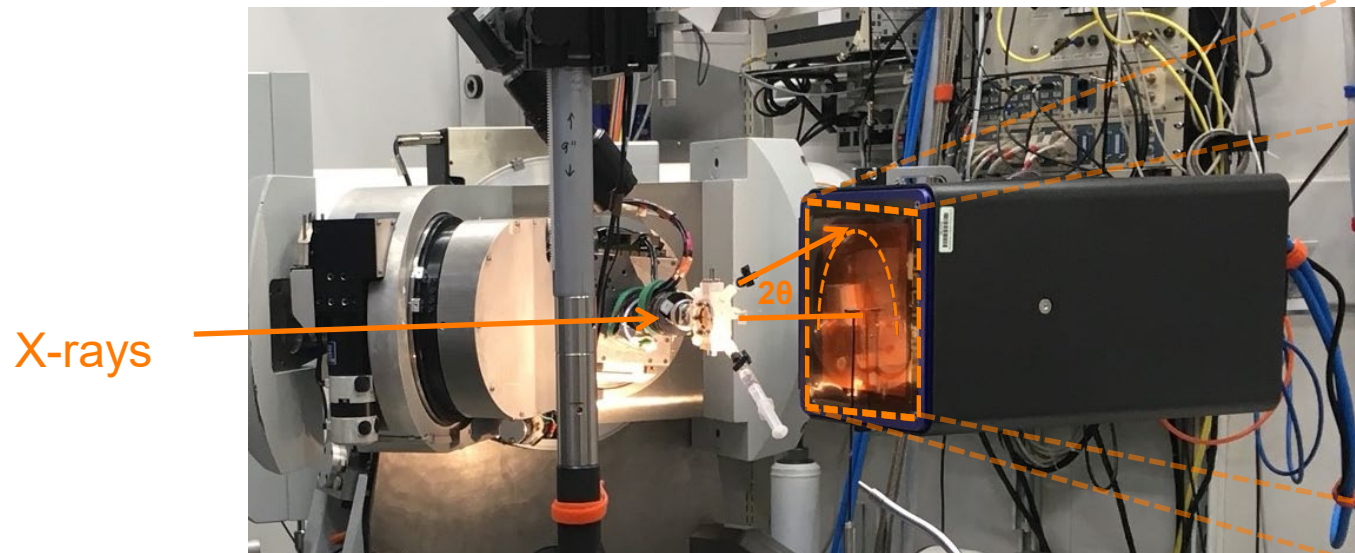
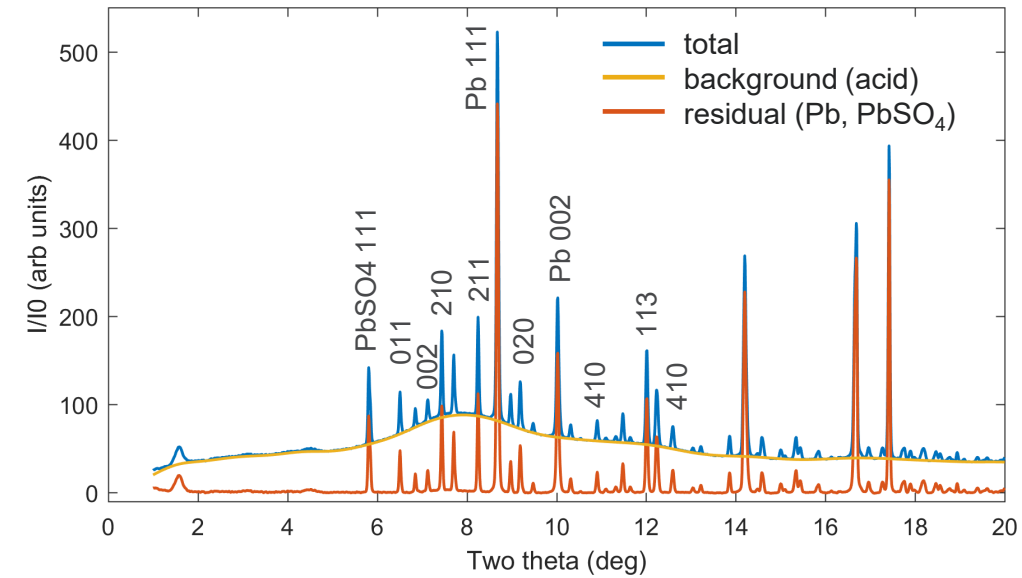
Detector (Pilatus 1M)

Potentiostat

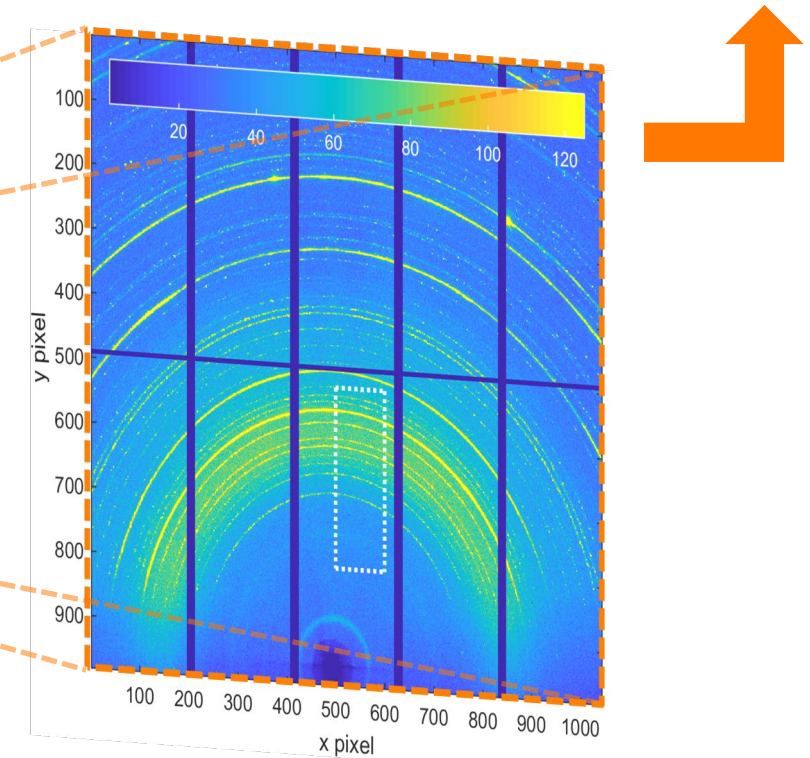
EXPERIMENT

Scattering from a lead surface

- A single detector image contains a wealth of information.
- Bragg peaks from active material phases (Pb, PbSO₄)
- Background from scattering from acid



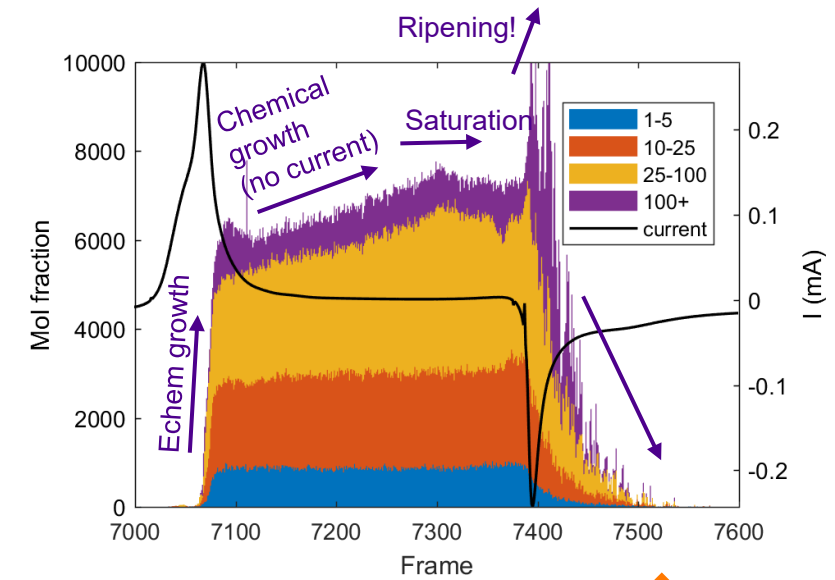
APS 13BM-C



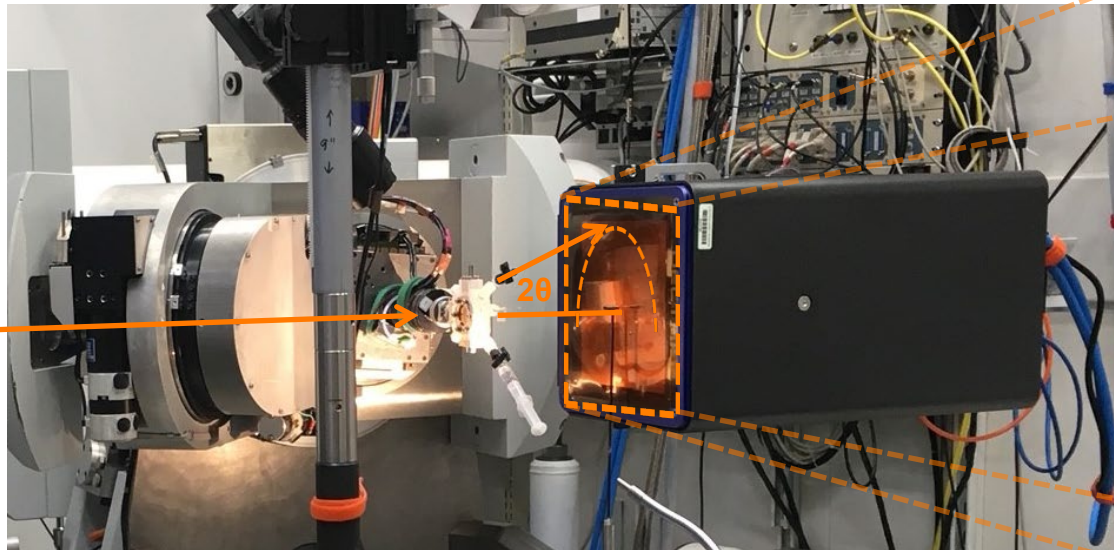
EXPERIMENT

Scattering from a lead surface

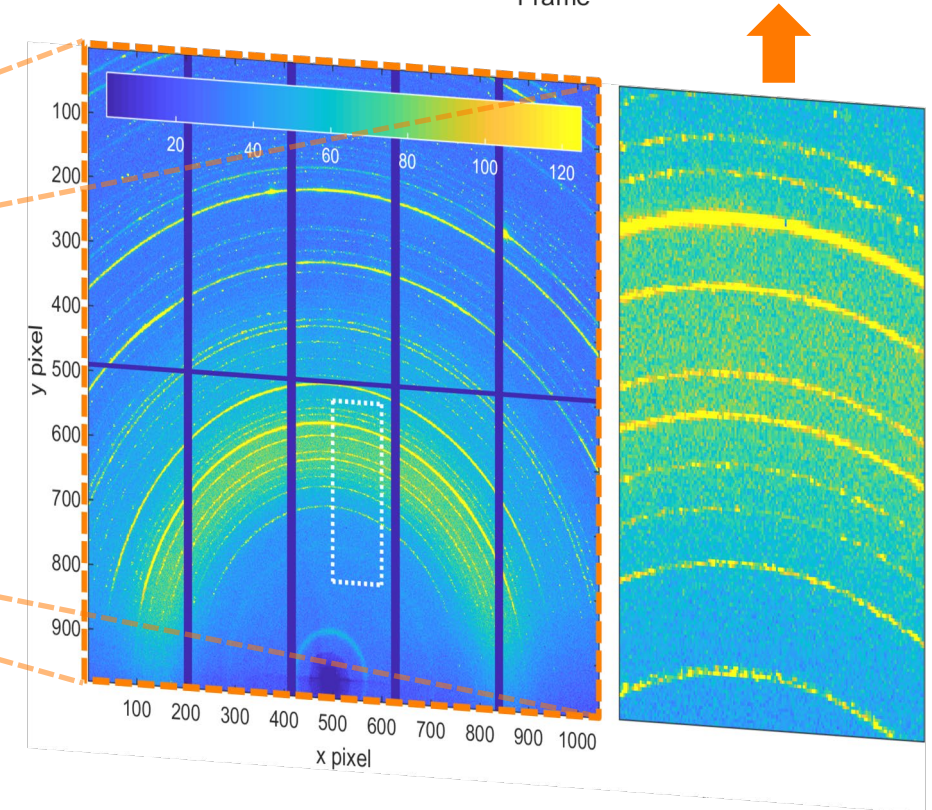
- A single detector image contains a wealth of information.
- Bragg peaks from active material phases (Pb, PbSO₄)
- Background from scattering from acid
- Powder lines = scattering from many crystals = statistics!



X-rays



APS 13BM-C

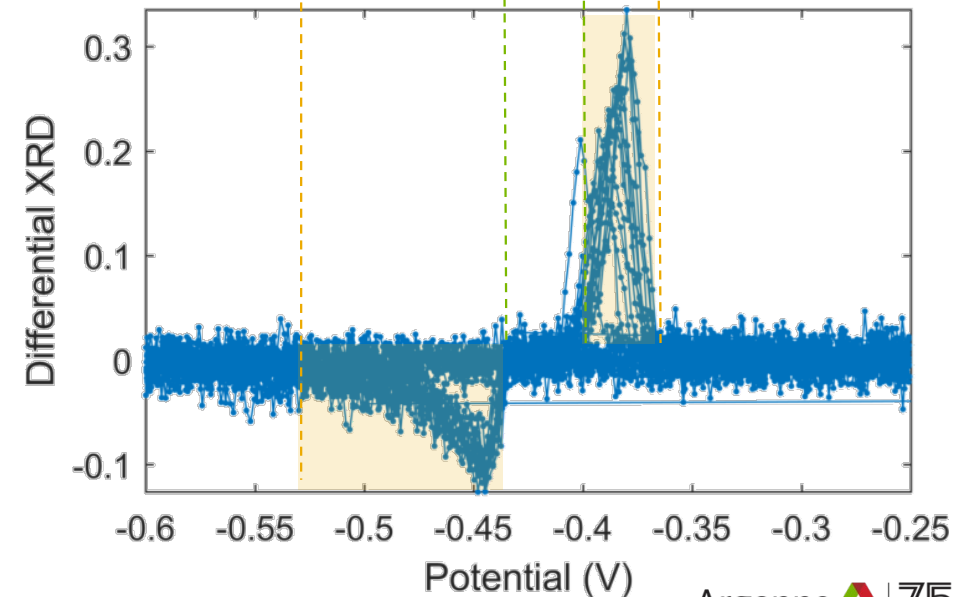
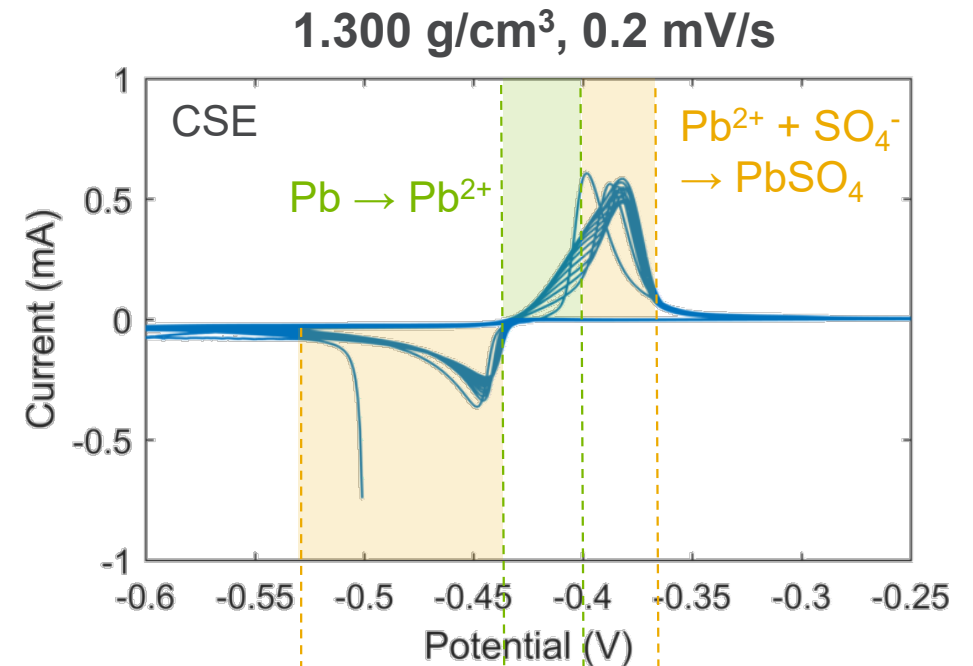
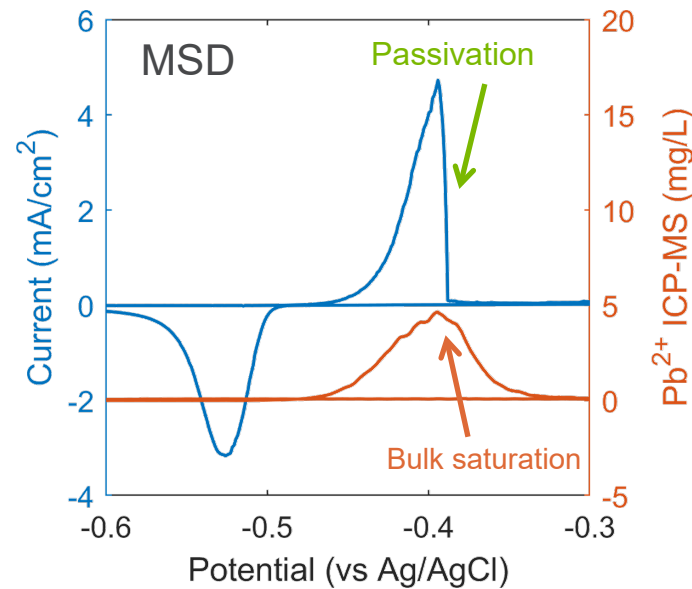


CV & XRD

Onset, kinetics

Comparison of current response versus *change* in PbSO_4 signal.

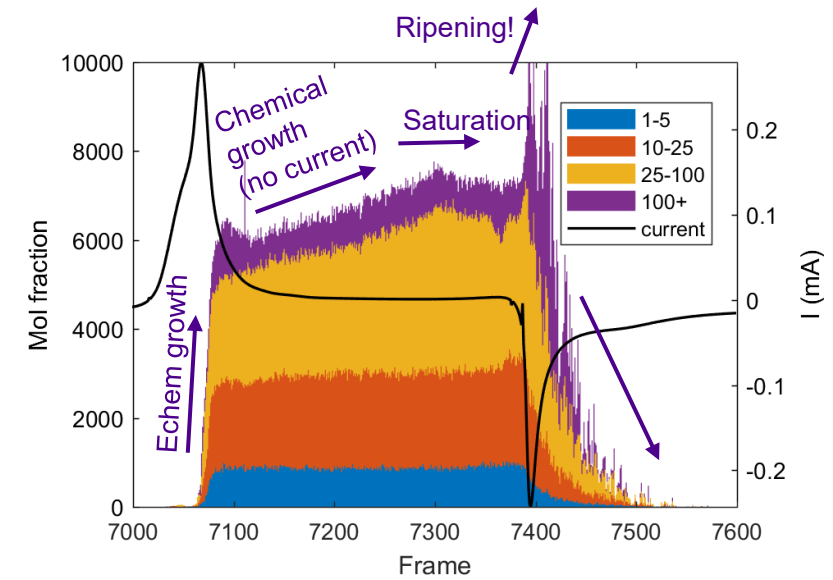
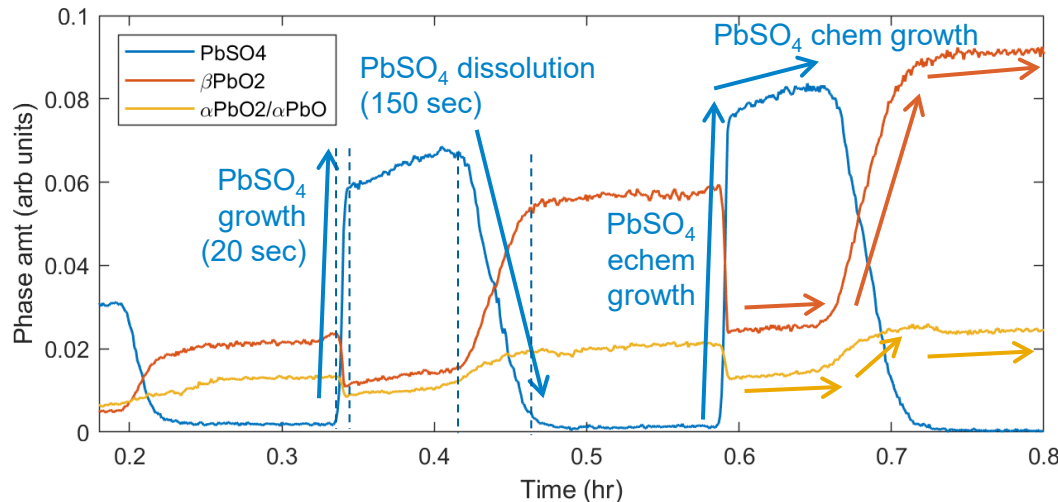
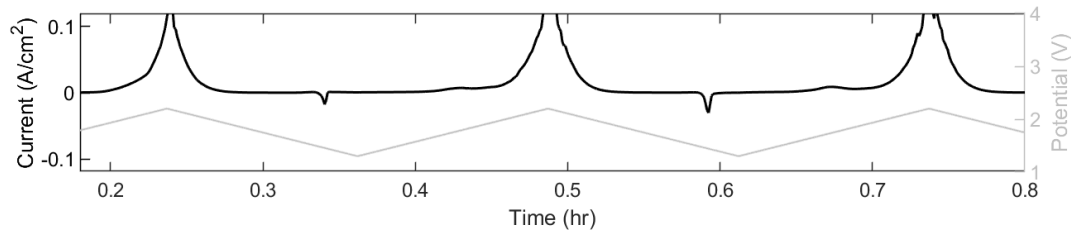
- Discharge: PbSO_4 growth delayed from onset of current and is more abrupt, suggesting a two step process.
 - Fits into analysis by MSD using in situ ICP.
- Charge: PbSO_4 dissolution rate similar to cathodic current \rightarrow crystal dissolution is the rate limiting step.



PLANTE POSITIVE

PbO, PbO₂, PbSO₄

- Cycles 1-3: discharge faster than charge, but charge process is fairly linear.
 - Note that PbO₂ dissolution is not reversible



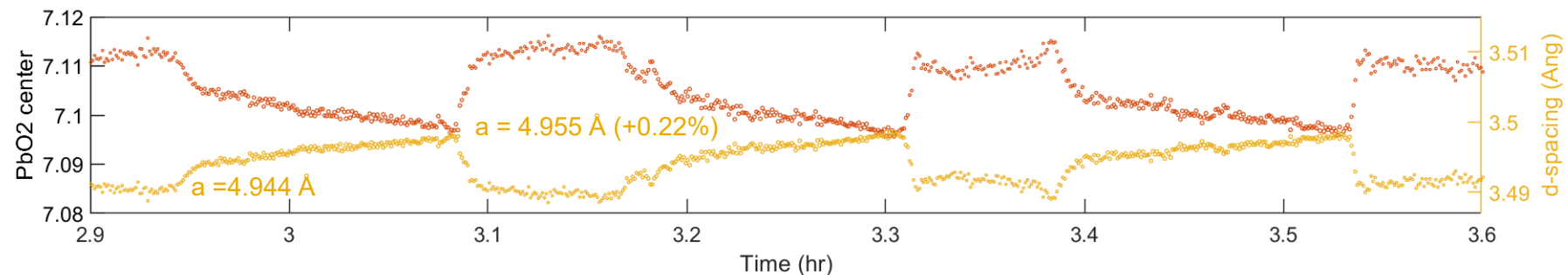
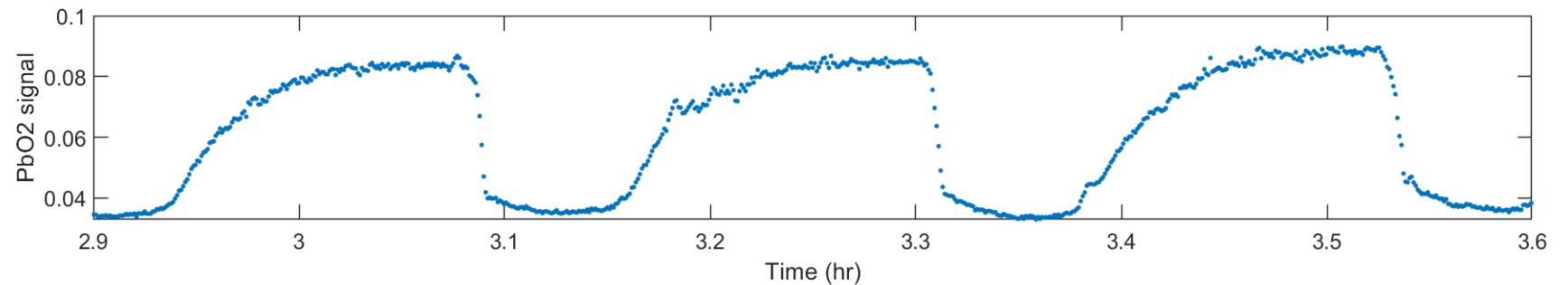
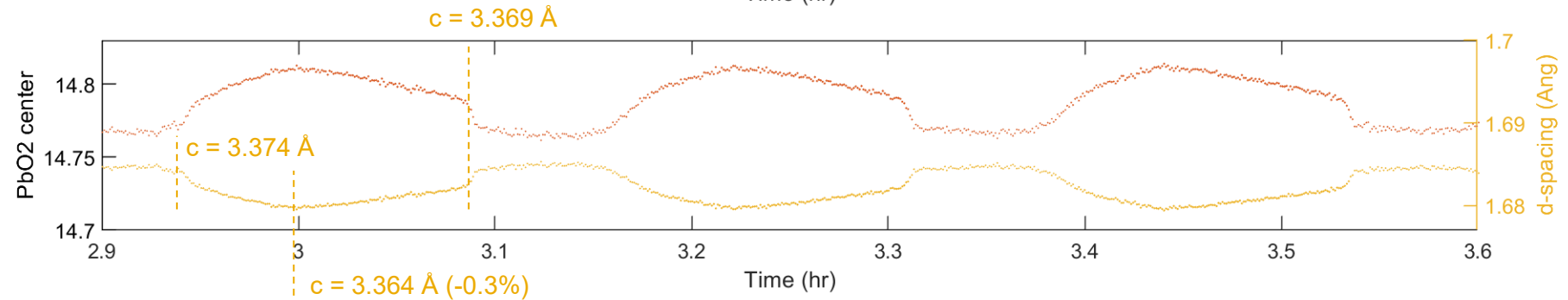
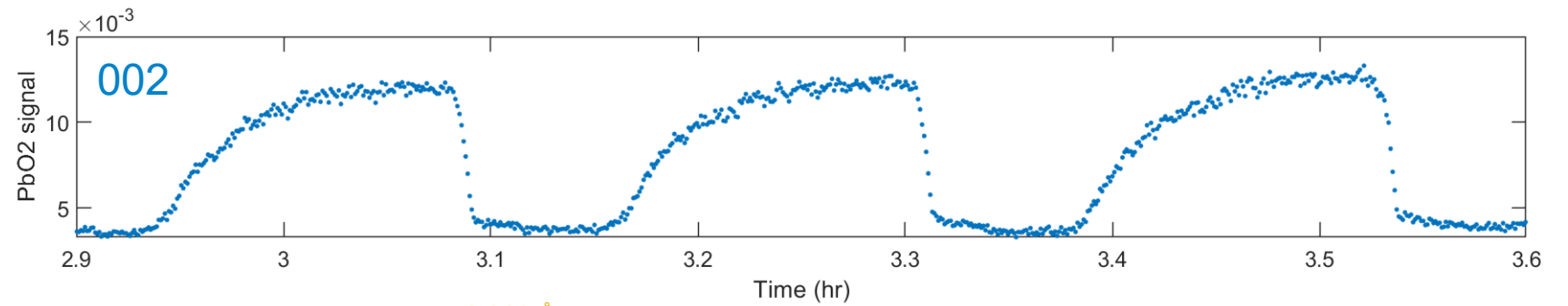
PbSO₄ growth/dissolution is similar to negative electrode (figure from review presentation and 12/2018 and 2/2019 talks):

- 1) Rapid growth during cathodic wave
- 2) Slow growth after (chemical growth of PbSO₄ as Pb²⁺ concentration reaches equilibrium).

Note that PbO/PbO₂ also grows at this condition!

PLANTE POSITIVE Peak shifts

- The crystal structure of each phase changes significantly during cycling.
- This is related to oxygen non-stoichiometry in PbO_2 .
- Overall volume change of 0.3% (that's a lot!)

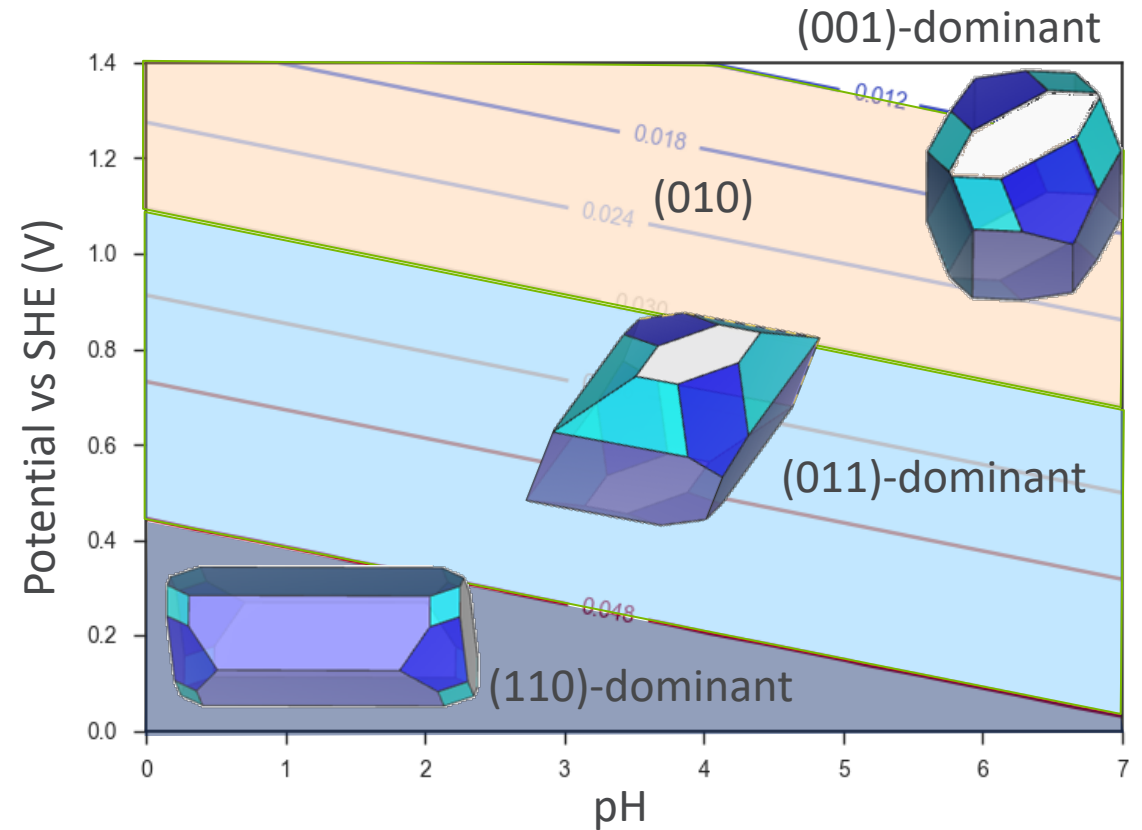
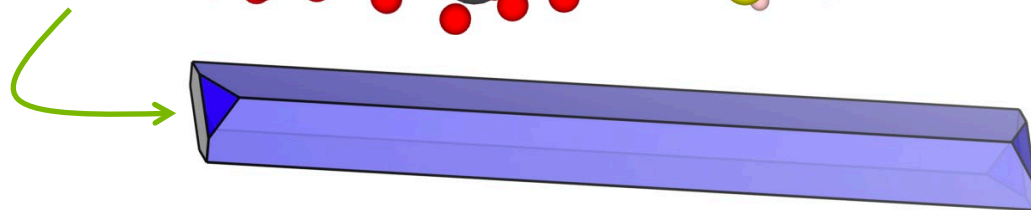
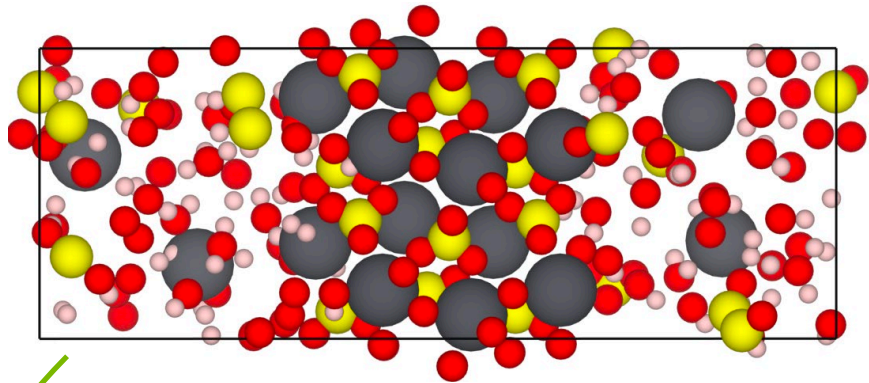


CRYSTAL HABIT

DFT/MD Approach

Big question: what is the most stable (or most reactive) PbSO_4 surface?

- DFT calculations of low-index PbSO_4 surfaces: equilibrium crystal shape
- Right: additional calculations were used to extend this approach to full range in pH.



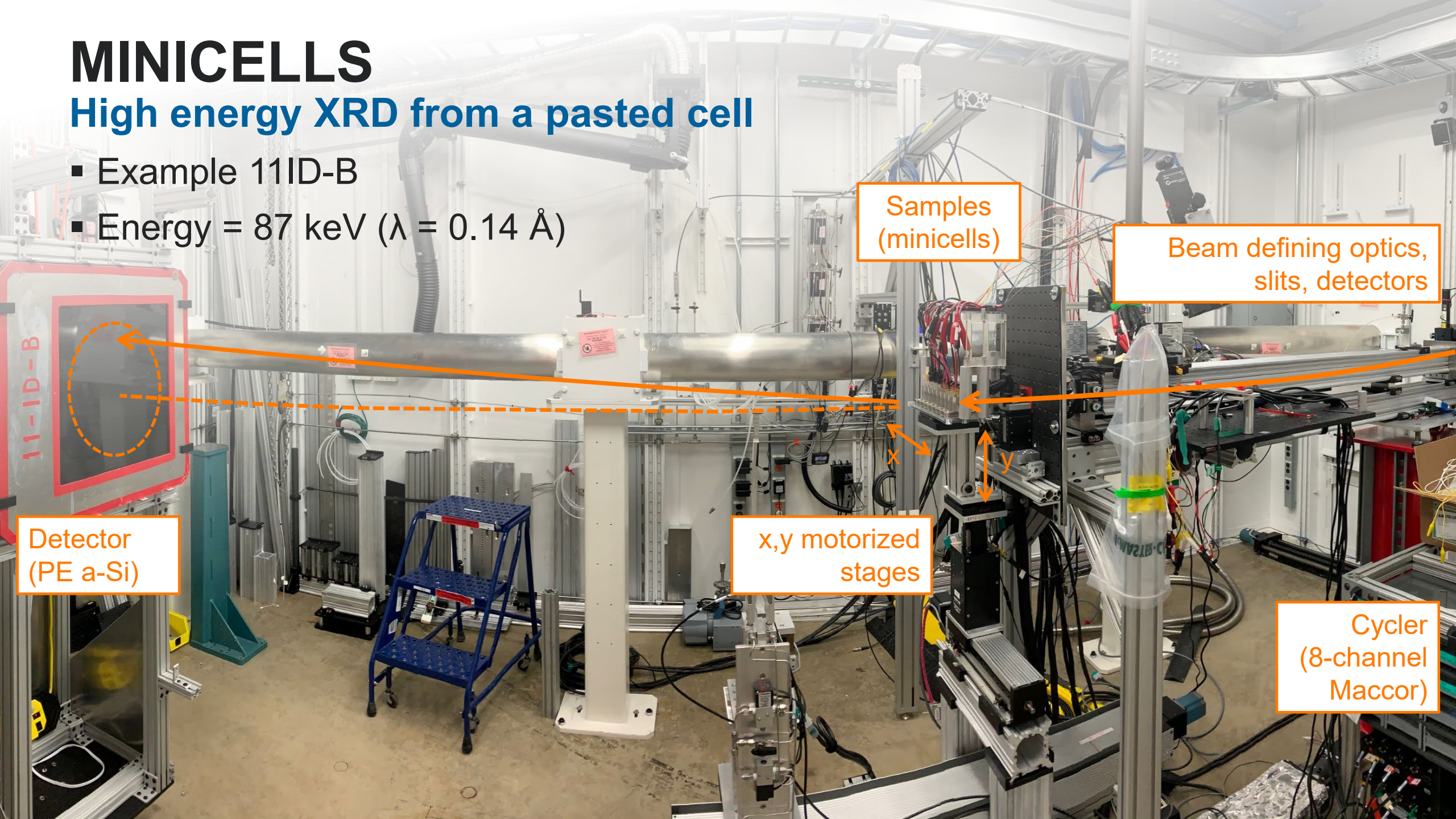
However, crystal growth/dissolution may not have time to equilibrate during charge and discharge. Ab initio MD (left) was used to study kinetically-limited processes, which have different looking crystal habits.

Can we use this information to **design** PbSO_4 crystals (or BaSO_4) with faster dissolution kinetics?

MINICELLS

High energy XRD from a pasted cell

- Example 11ID-B
- Energy = 87 keV ($\lambda = 0.14 \text{ \AA}$)



Samples
(minicells)

Beam defining optics,
slits, detectors

Detector
(PE a-Si)

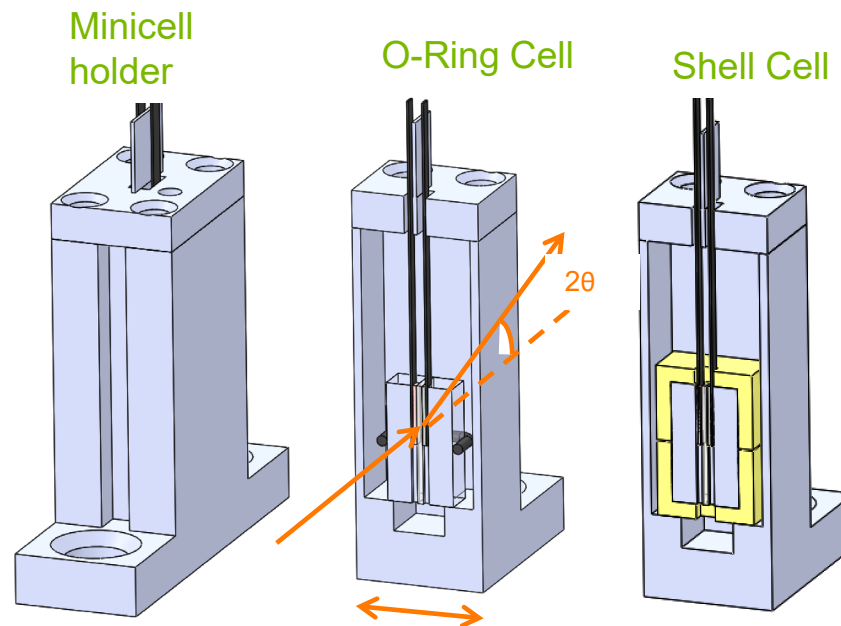
x,y motorized
stages

Cycler
(8-channel
Maccor)

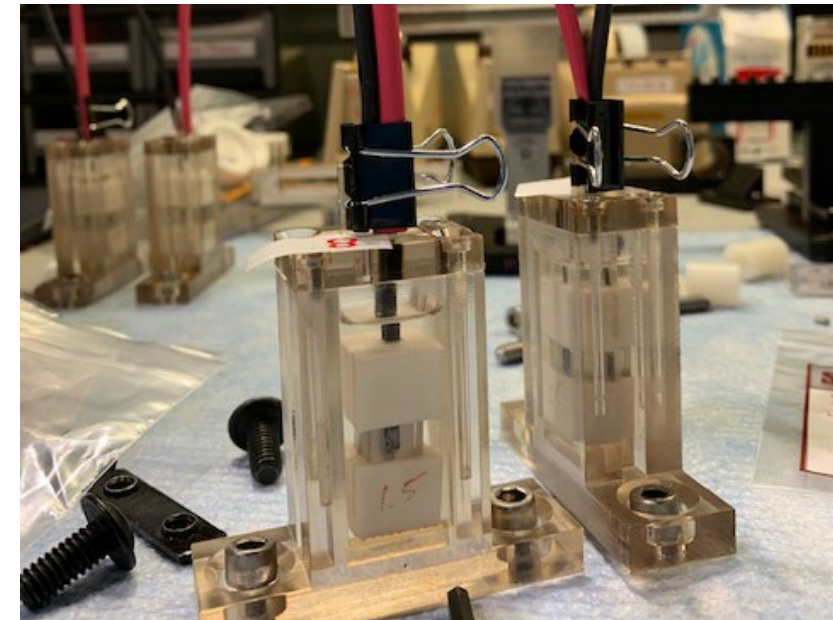
MINICELLS

Construction

- Minicells developed at East Penn for x-ray depth profiling.
- Initial cells used O-ring(s) to define compression, but later designs used an external shell.
- Parts were printed or machined from acrylic.



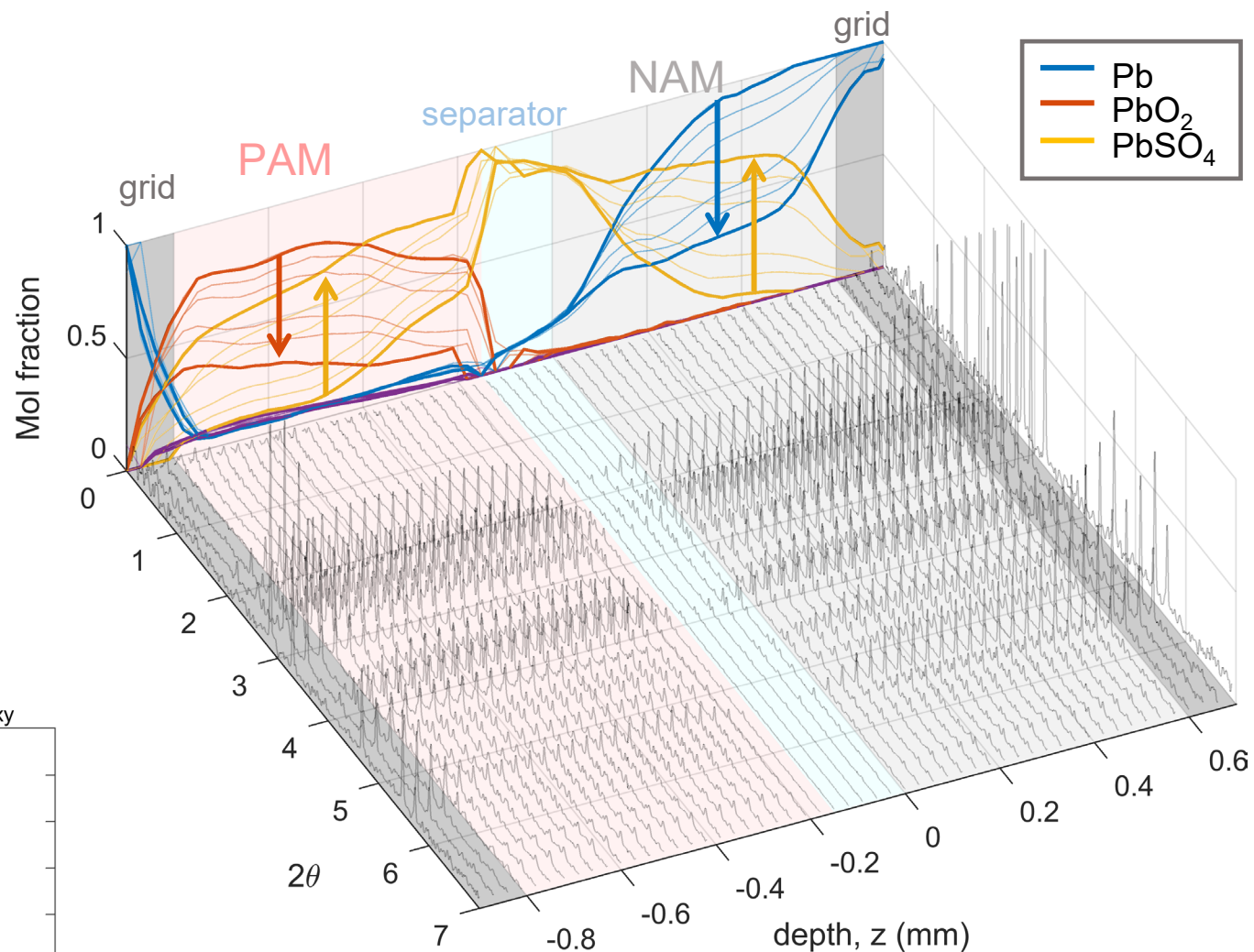
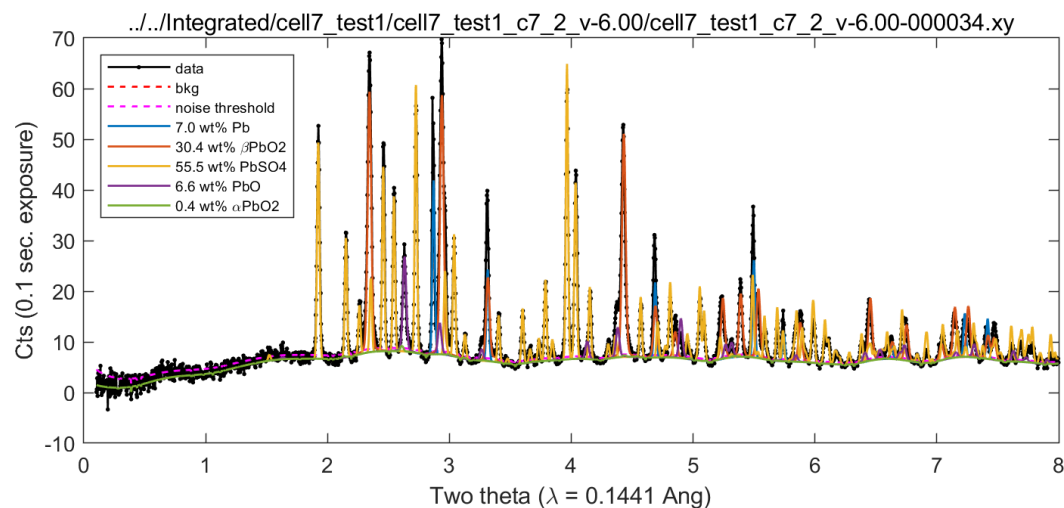
Depth, $z = \pm 1.5 \text{ mm}$
(100 steps: $30 \mu\text{m}/\text{step}$)



DEPTH PROFILING

Example

- Typical step size of 0.03 mm through the cell with 0.1 sec integration time.
- Example of data from PAM, below.

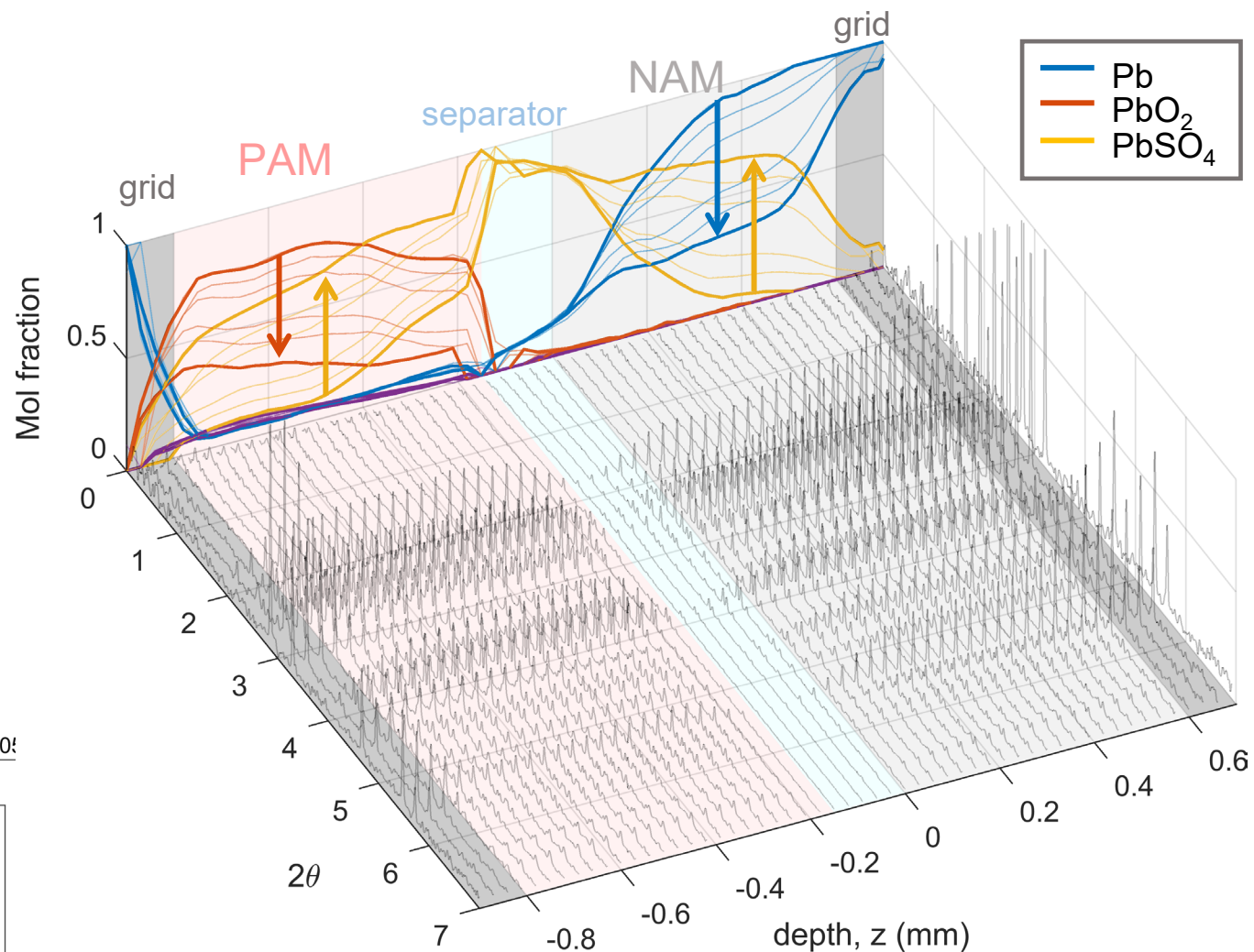
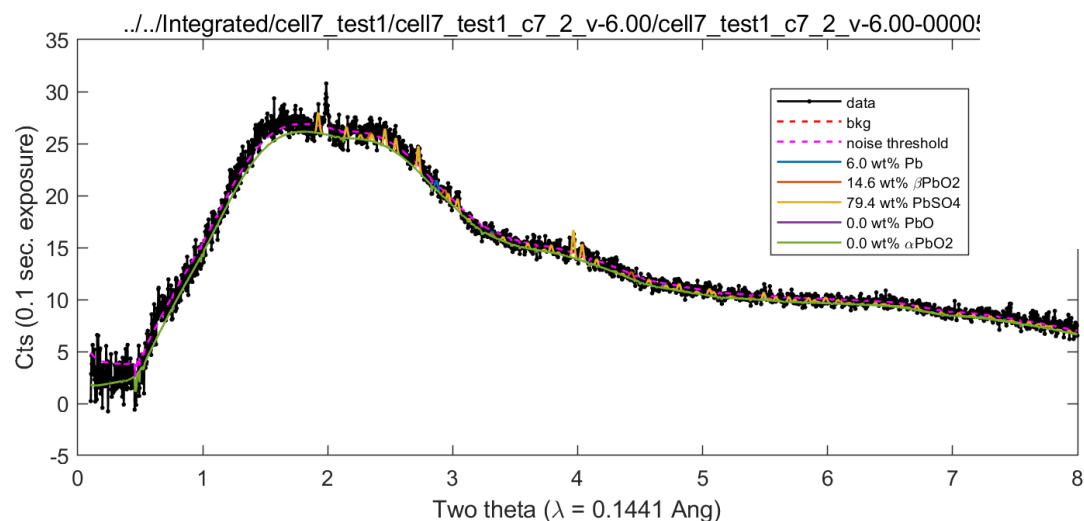


Important: minicells get 60-80% utilization at 1C rates, yet macrocells/batteries get 20-30% utilization at 1C.

DEPTH PROFILING

Example

- Typical step size of 0.03 mm through the cell with 0.1 sec integration time.
- Example of data from separator region, below:

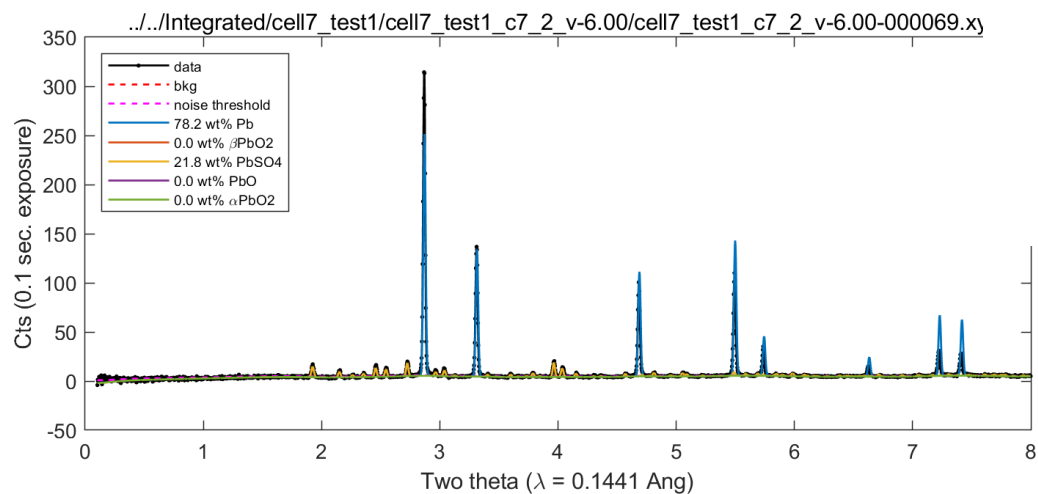
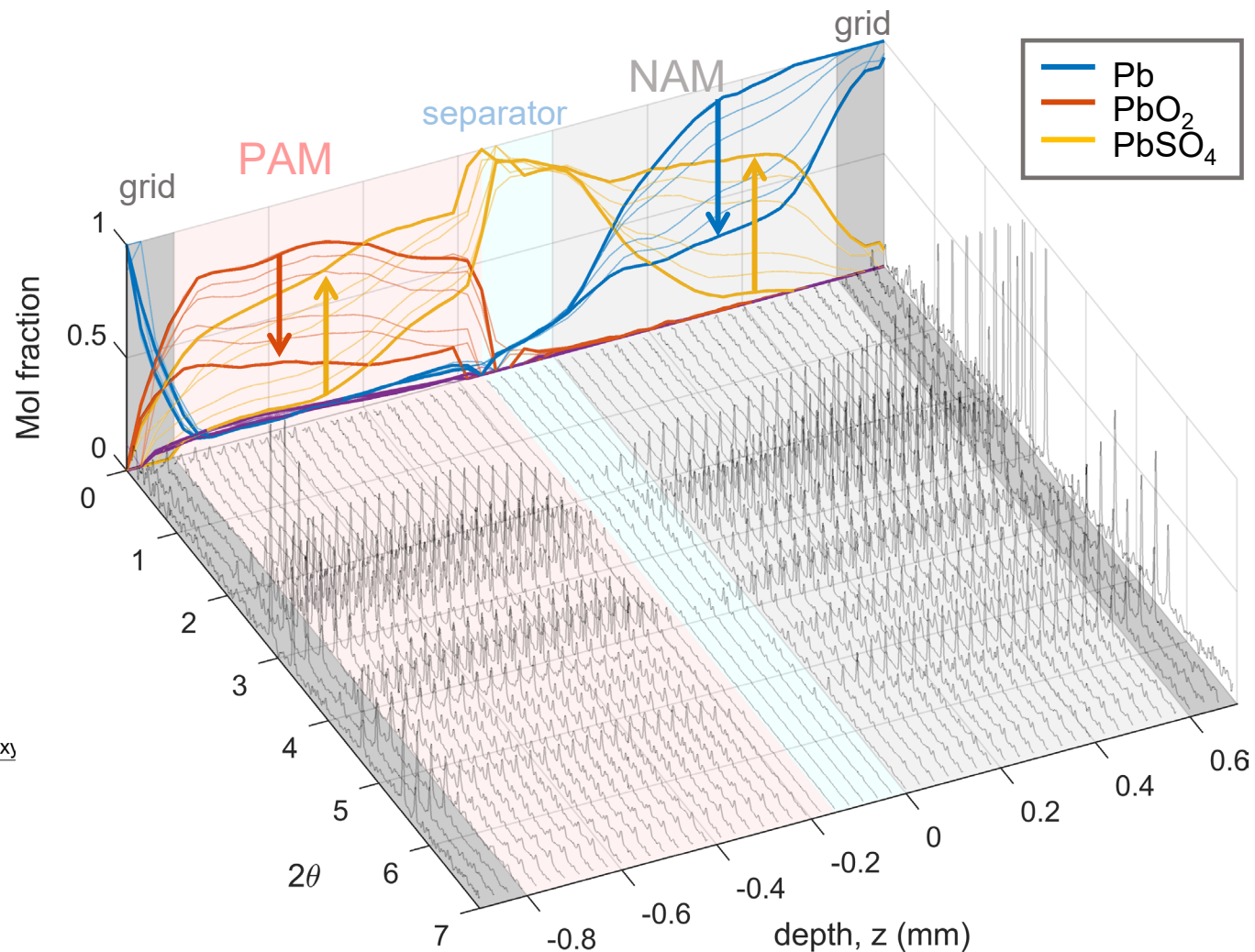


Important: minicells get **60-80 %** utilization at 1C rates, yet macrocells/batteries get 20-30% utilization at 1C.

DEPTH PROFILING

Example

- Typical step size of 0.03 mm through the cell with 0.1 sec integration time.
- Example of data from NAM, below.

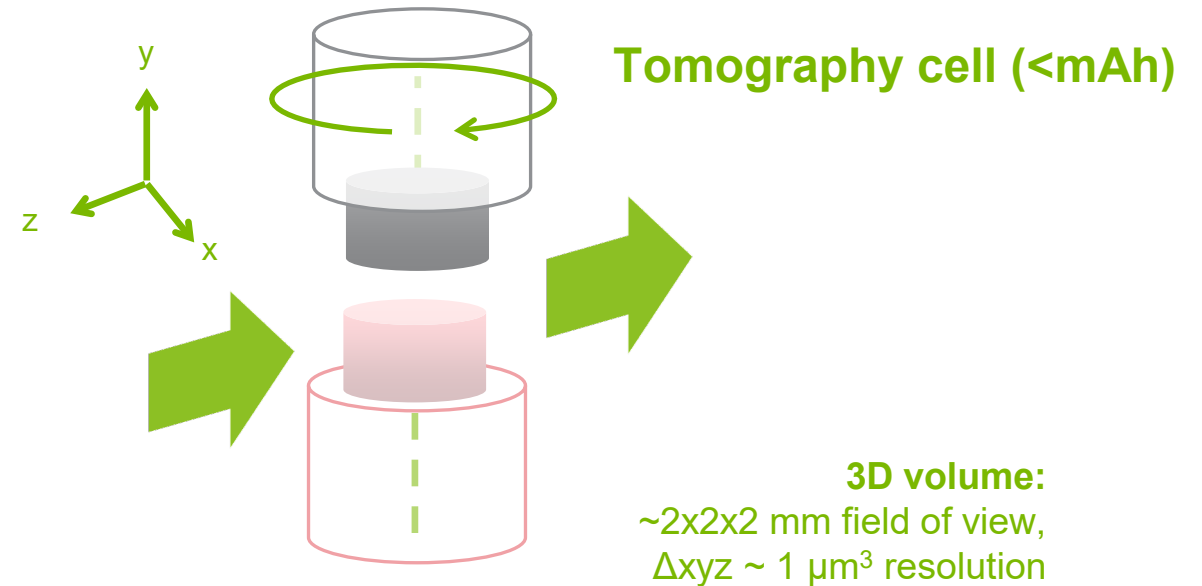
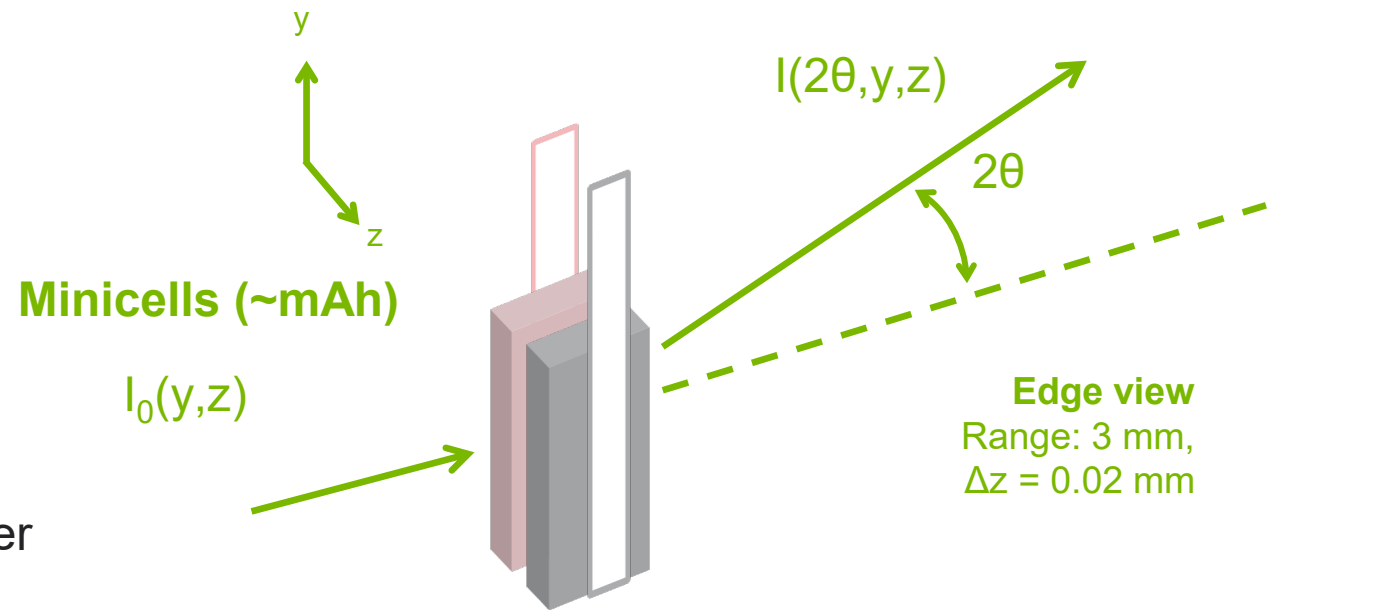


Important: minicells get **60-80 %** utilization at 1C rates, yet macrocells/batteries get 20-30% utilization at 1C.

MINICELLS FOR TOMOGRAPHY

Cylindrical geometry

- Unlike 1D depth profiling, computed tomography requires ~1000 projections over 180° rotation.
- Minicells were explored for CT studies, but suffered from artifacts from the highly absorbing lead tabs on the sides.
- Wolfman developed a cylindrical cell, with top and bottom electrodes that provided much better reconstructions. Pasting/curing is still a challenge.
- Recent upgrades at sector 7BM have incorporated energy dispersive diffraction as a secondary probe.

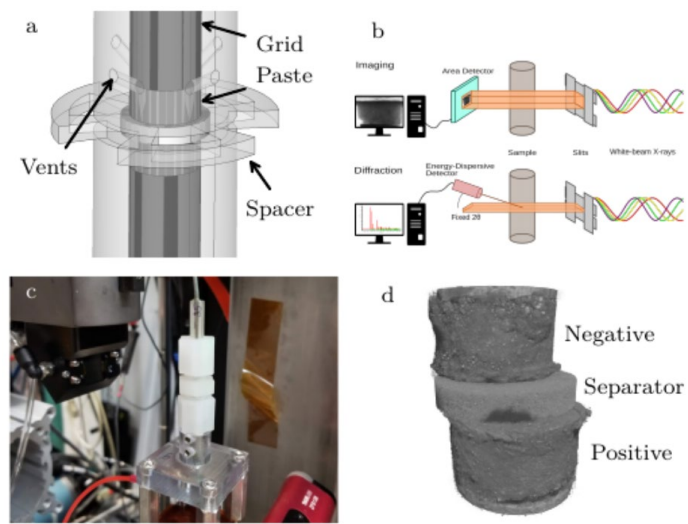


1000-2000 projections:
3D reconstruction

MICRO-CT

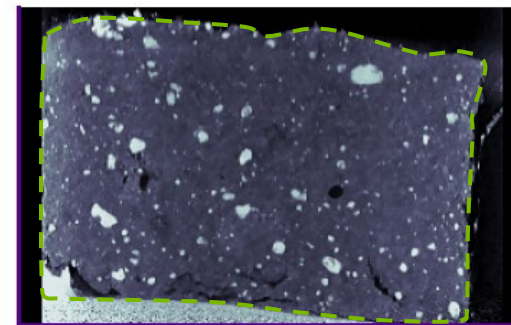
Imaging a whole electrode

- Biggest changes in porosity found during formation, especially on negative.
- These voids lead to densification of the active material.

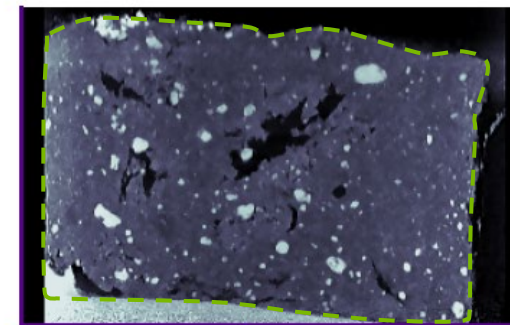


NAM cross-section during formation and cycling

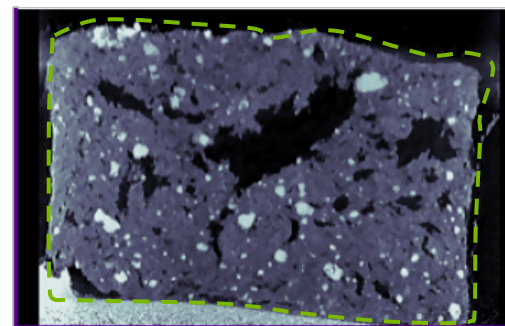
Before Formation



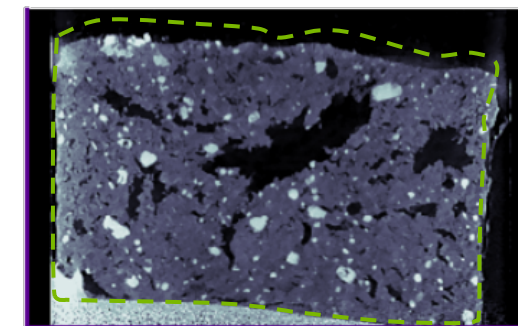
2h Formation



9h Formation



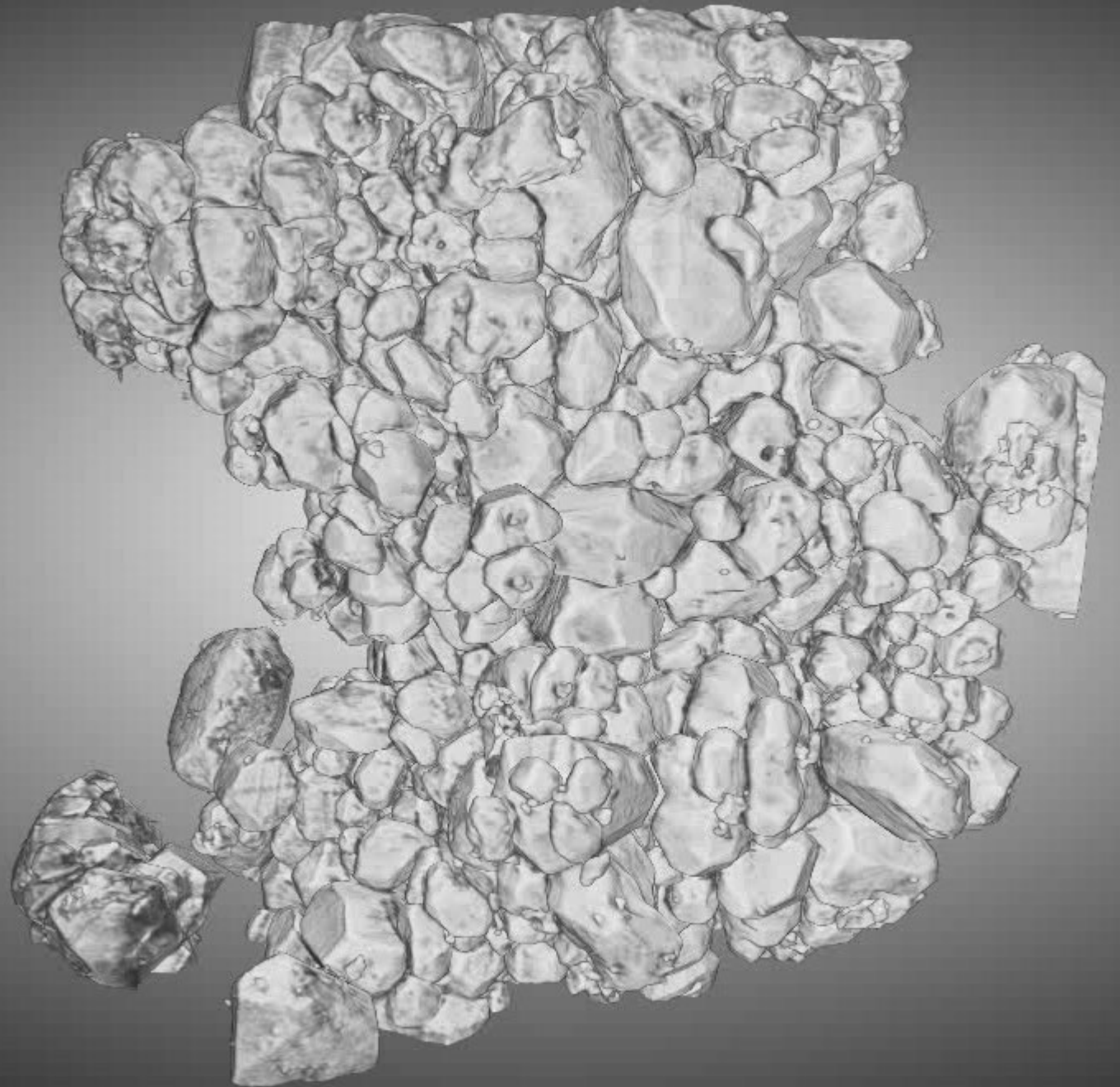
After Discharge



NANO-CT

3D microstructure

- Transmission x-ray microscopy (TXM) reconstruction on end-of-life negative active material (40 nm resolution).
- Reconstruction: can isolate species, including pores and sponge lead.
 - Lead network crucial for utilization, rechargeability.
- Understanding surface area, pore distribution, tortuosity, etc. important for modeling these processes.



MACROCELLS

XRD maps

- 2V cells developed at Clarios and Gridtential.
- Protocols: 80% DOD cycling and HRPSOC.

Conventional cell
1N/1P ~1.3Ah
(Clarios)

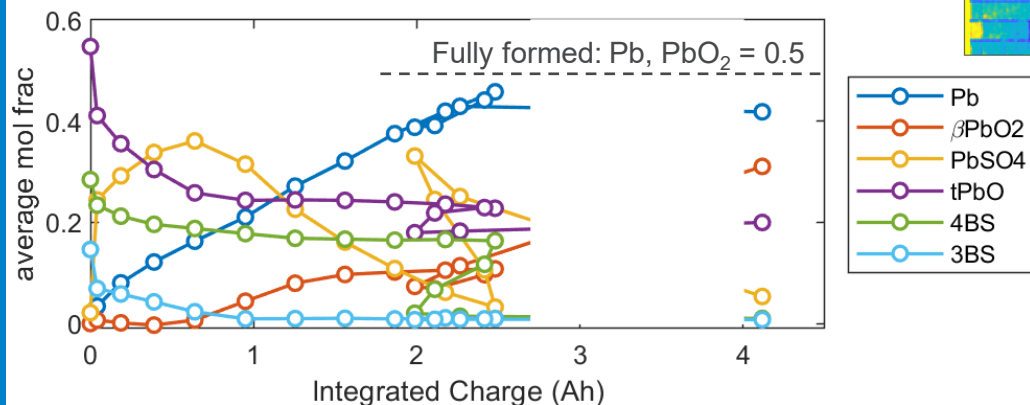
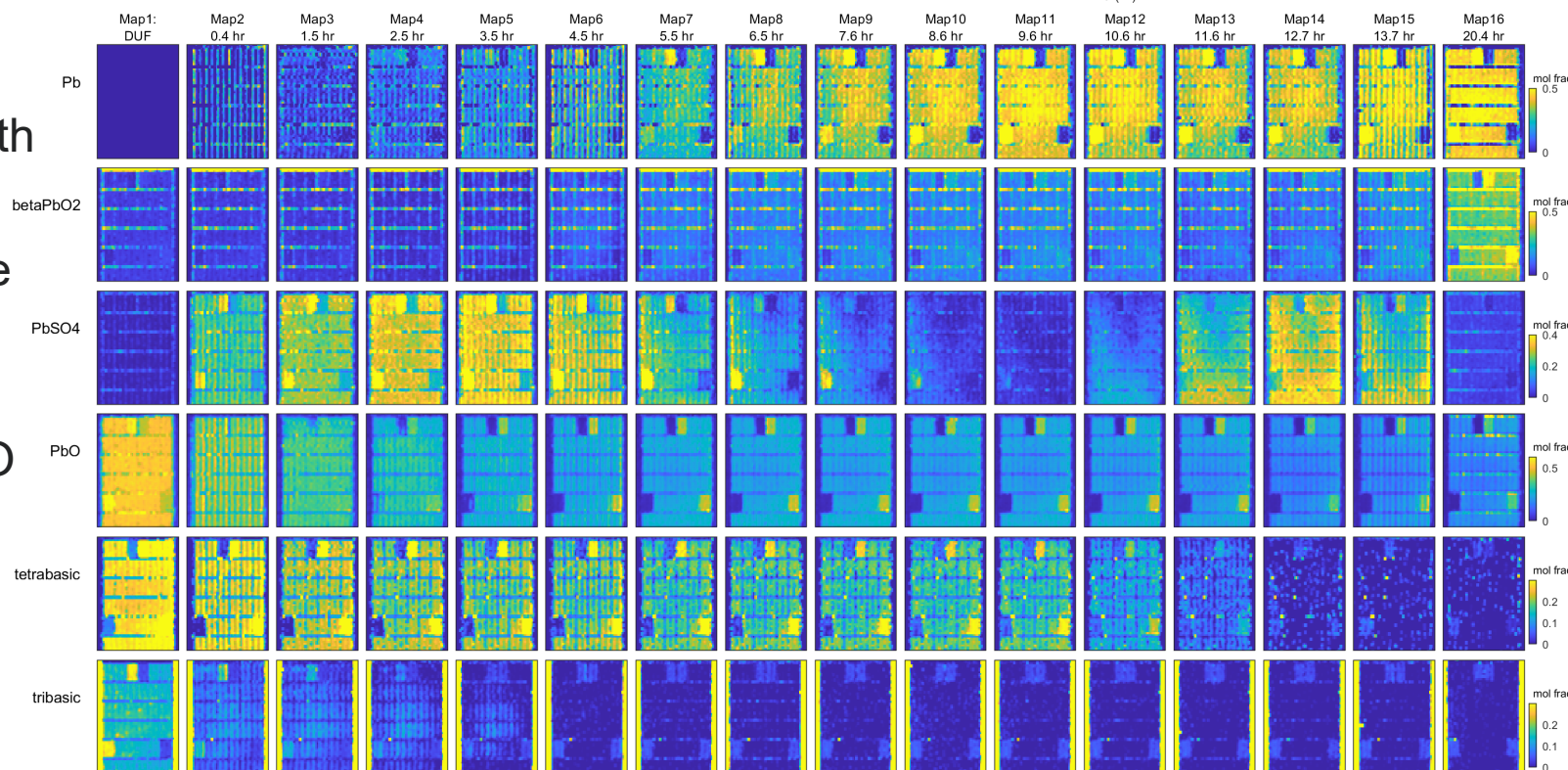
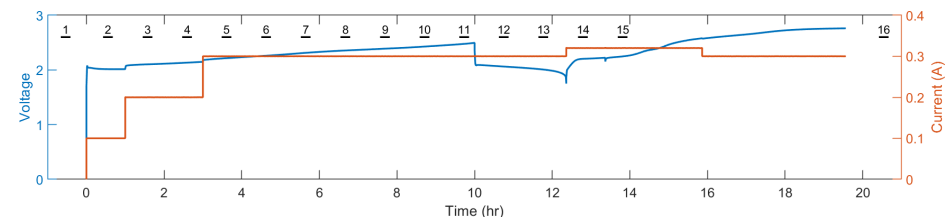
Bipolar cell
1N/1P 3-4 Ah
(Gridtential)



FORMATION

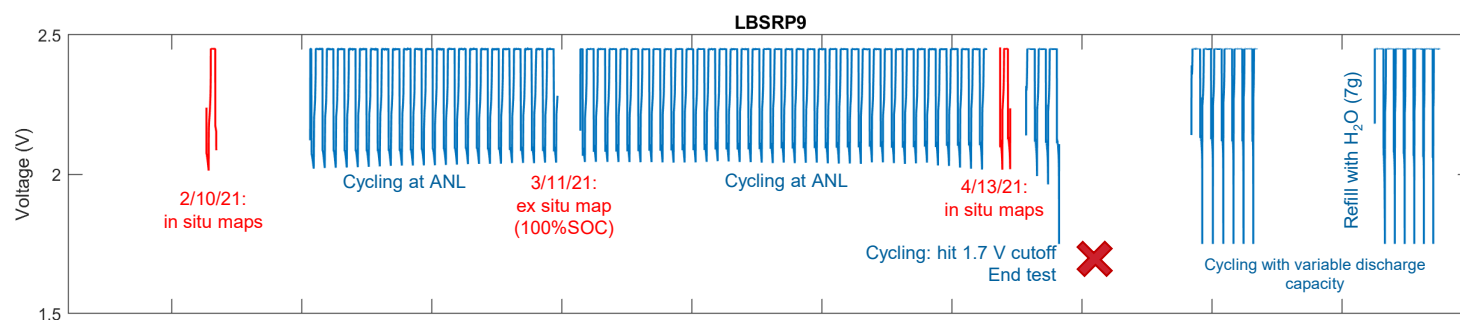
In situ diffraction maps

- Example: formation on gen4 cell with 4BS positive and 3BS negative.
- See variation in PbSO_4 and alkaline phases, driven by accessibility to acid.
- 4BS lasts longer than 3BS, but PbO is present at the end of formation.
- Windows show that PbO is present only on positive.



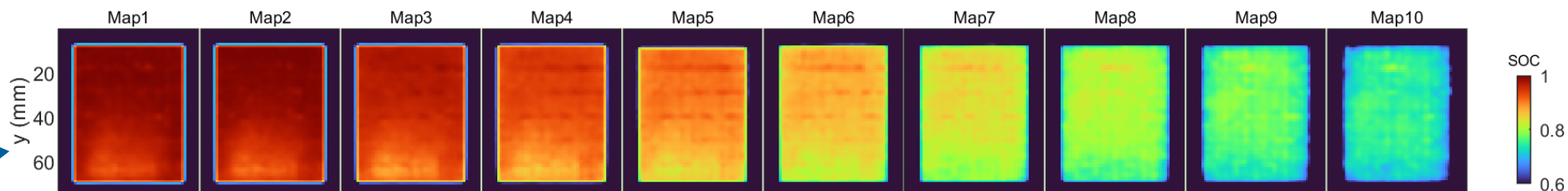
CYCLING

Conventional cell

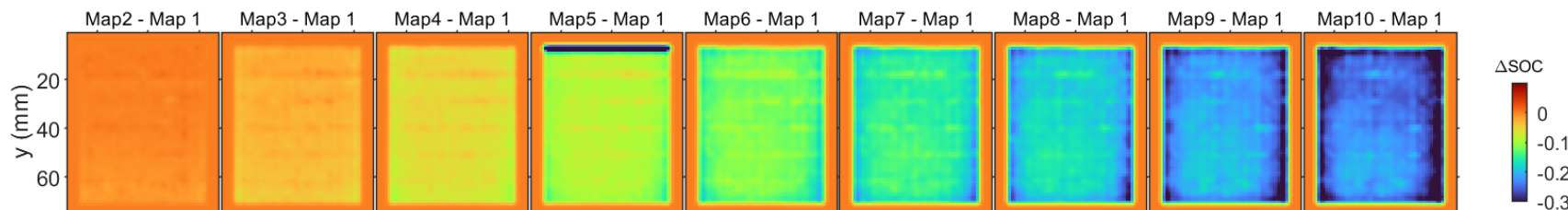


Example from baseline cell. Maps:

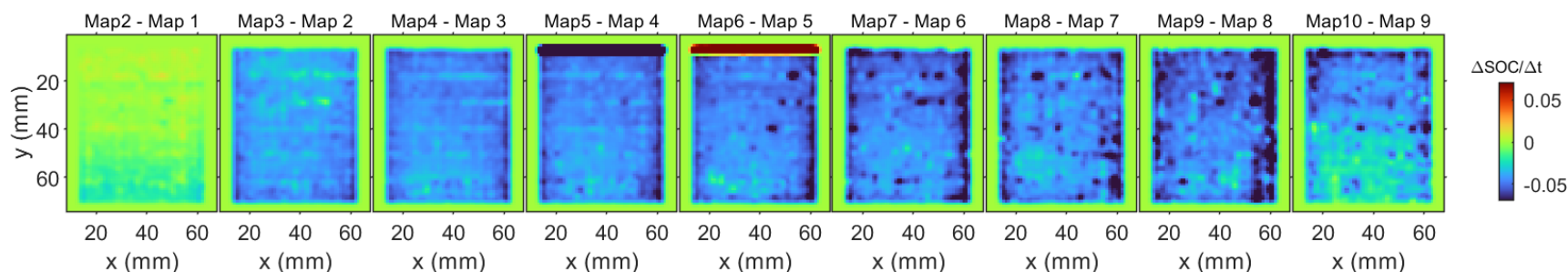
- Overall SOC



- Change in SOC from $t = 0$ = utilization



- $\Delta SOC / \Delta t$ ~ current density



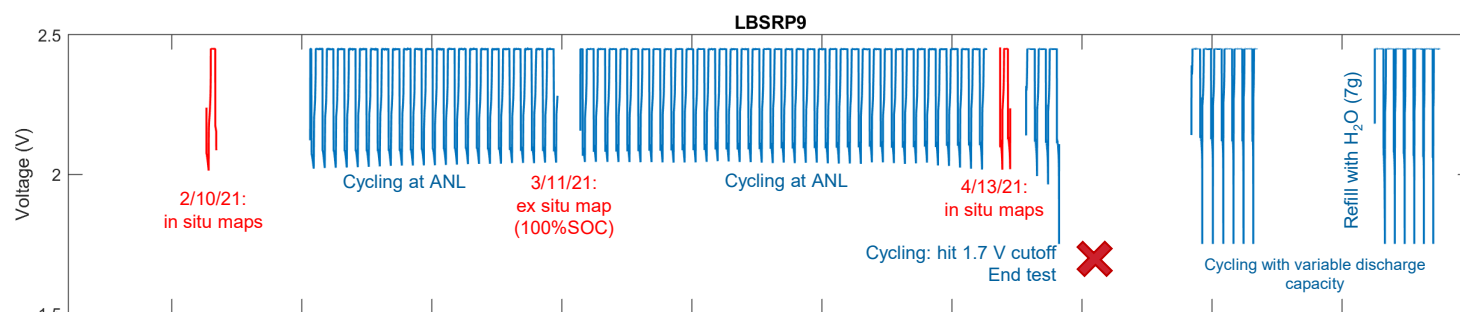
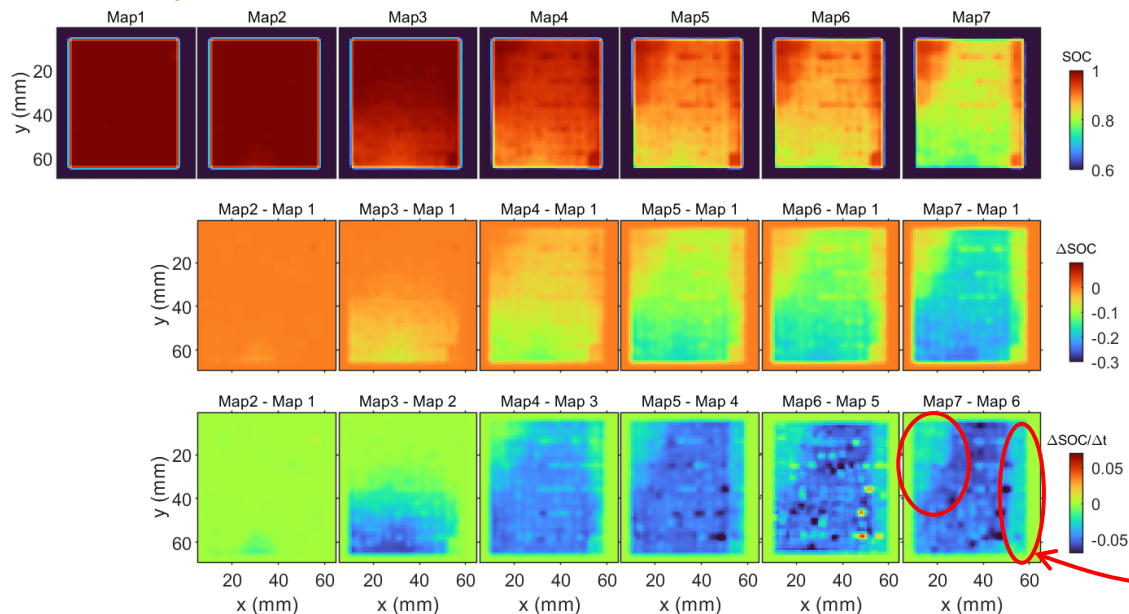
Initial cycling shows largely uniform charge acceptance.

CYCLING

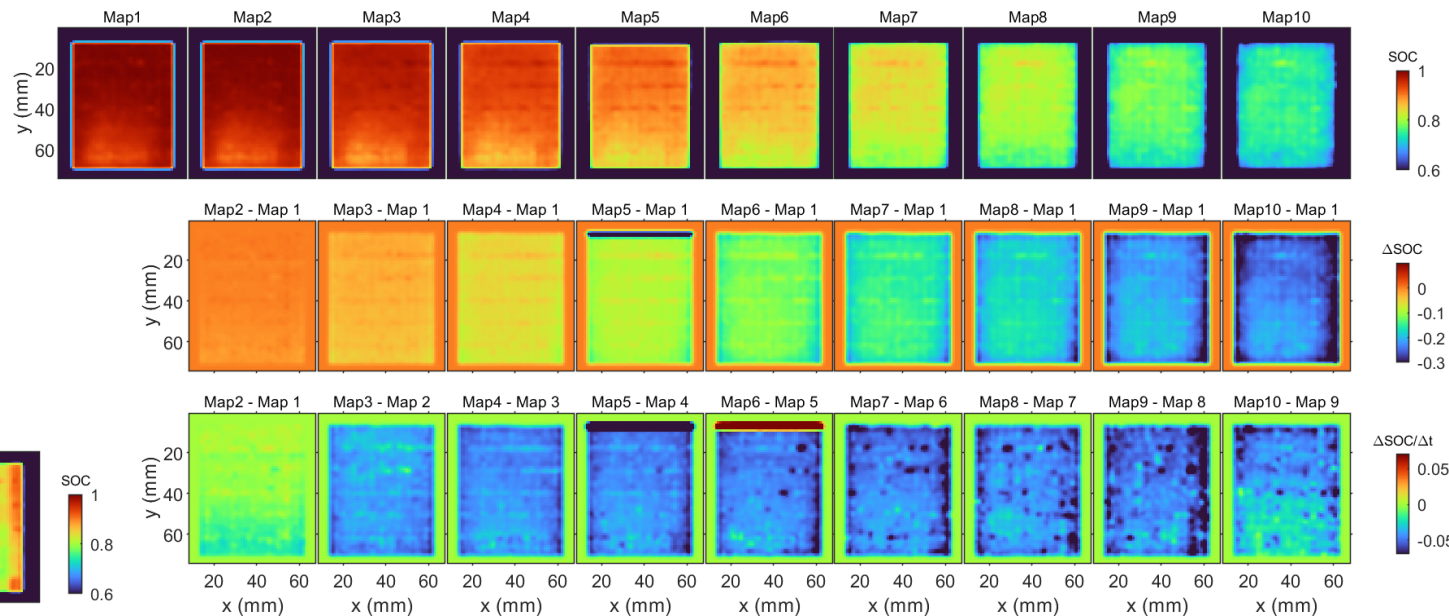
Conventional

- Initial cycling (2/21 experiment) showed uniform charge acceptance during discharge, apart from PbSO_4 accumulation around periphery at high DOD.

4/21 experiment



2/21 experiment



- After 2 months of deep cycling, utilization is much more nonuniform. This driven by the PAM, which does not form PbSO_4 during discharge in these problem regions.

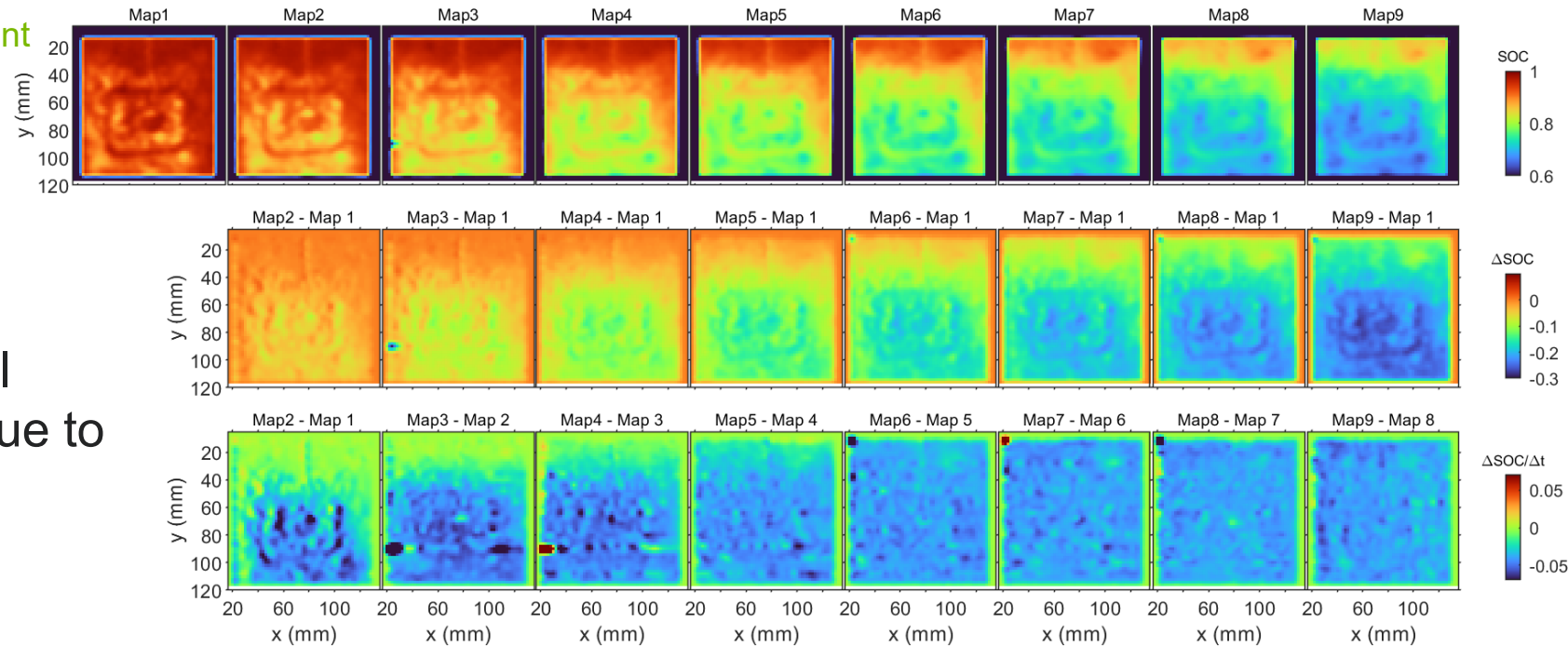
Inactive regions

CYCLING

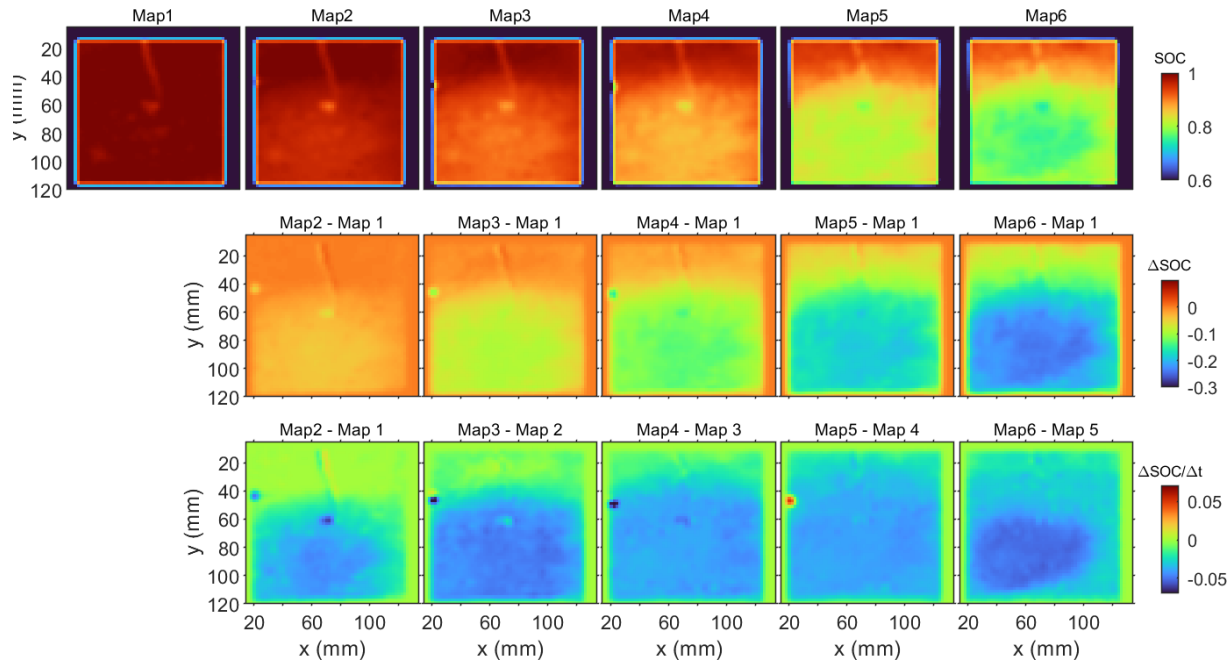
Bipolar

- Data from our first bipolar cell shows heterogeneity, likely due to vacuum filling at first.

2/21 experiment



4/21 experiment



- After two months cycling, the bipolar cell has improved uniformity (apart from inactive region near the top).
- Even the current density is highly uniform compared to earlier cycling and monopolar cells.
- Cycle life was ultimately limited by 0.1 mm lead layer, which has been increased to 0.4 mm.



SUMMARY AND FUTURE



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LEAD ACID AT NATIONAL LABS

Current work

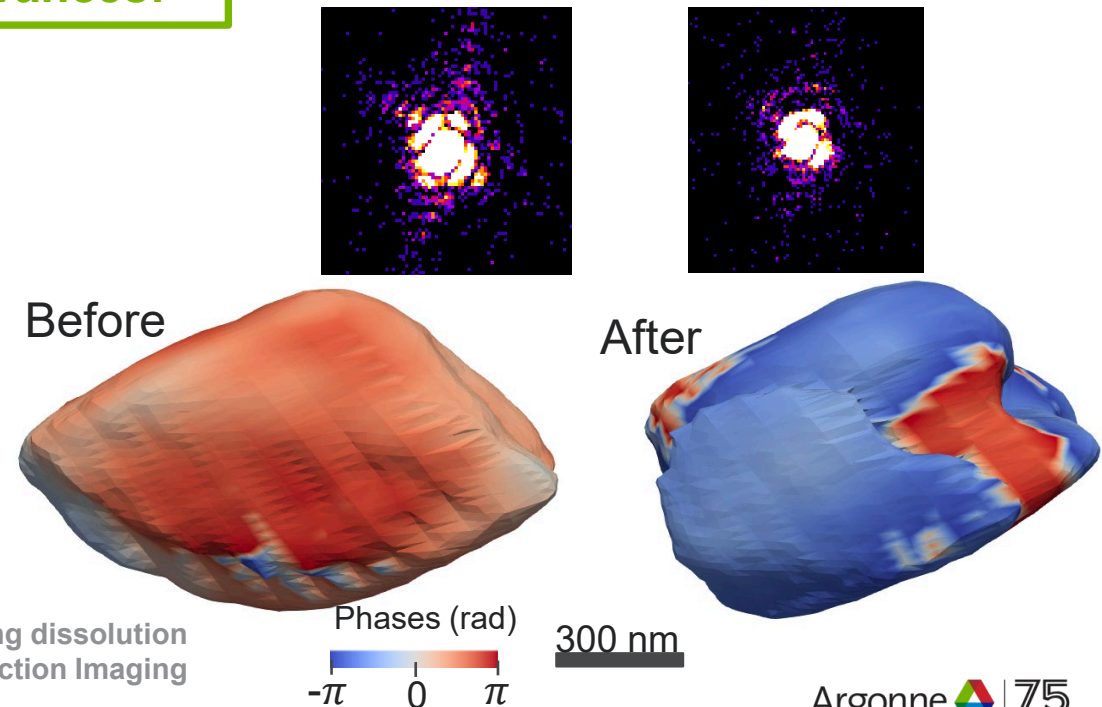
- There is still a tremendous amount to learn about the science of lead acid batteries.
- There is still room for lead acid to improve in utilization and cycle life.

Communication between national labs and industry is crucial to make advances!



Future work

- APS is undergoing a \$815M upgrade that will revolutionize this approach.
- Smaller, coherent beams: isolate and watch change in size and shape of individual crystallites



Data collected on synthetic PbSO_4 during dissolution using Bragg Coherent Diffraction Imaging

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